Crystal Structure of a Novel Furo-Furan Lactone from *Heliotropium eichwaldi*

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A novel furo-furan lactone **1** has been isolated as fine needles from the ethyl acetate-soluble fraction of *Heliotropium eichwaldi*. This is the first report of any furo-furan from the genus *Heliotropium*. β -Sitosterol 3-O- β -D-glucoside **2** has also been obtained for the first time from this source. The structure of **1** was elucidated by X-ray diffraction studies.

Key words: Heliotropium eichwaldi, Boraginaceae, Furo-Furan Lactone, X-Ray Diffraction Studies

Introduction

Heliotropium eichwaldi Steud (Boraginaceae) is an annual herb which grows in cultivated fields, gardens and shallow lands of the province of Sindh, Pakistan [1]. The plant is emetic, antidote and antiseptic. It is well known to be used for the treatment of scorpion stings, bee stings, mad dog bite, snake bite, earache, and cleaning and healing of wounds, warts and ulcers [1, 2]. Previous investigations on H. eichwaldi have yielded pyrrolizidine alkaloids and terpenoids [2]. The ethnopharmacological and chemotaxonomic importance of the genus Heliotropium prompted us to carry out further studies on H. eichwaldi. As a result, we now report the isolation and structural elucidation of a novel furo-furan lactone 1 along with β -sitosterol 3-O- β -D-glucoside (2), reported for the first time from this species.

Results and Discussion

The methanolic extract of the whole plant of *H. eichwaldi* was divided into *n*-hexane-, EtOAc-, *n*-BuOH- and water-soluble fractions. A series of column chromatography separations applied to the EtOAc fraction resulted the isolation of compounds **1** and **2**.

Compound 1 was obtained as fine colorless needles, m. p. 172 $^{\circ}$ C. The (+)-FAB HRMS showed a quasi

 $[M+H]^+$ peak at m/z = 275.1131 (calcd. 275.11308 for $C_{12}H_{19}O_7$). The IR spectrum showed the presence of an intense peak at 1799 cm⁻¹ corresponding to C=O vibrations and a broad peak in the region 3400-3200 cm⁻¹ suggesting the presence of OH functionalities. The molecular formula was confirmed by broadband (BB) and DEPT ¹³C NMR spectroscopy which showed twelve well resolved signals comprising of three methyl, one metylene, four methine and four quaternary carbon atoms. The most downfield signal at δ = 176.2 ppm could be assigned to the lactone carbonyl carbon atom. It further showed oxygen-containing quaternary carbons at δ = 80.1, 92.2 and 96.5, oxymethine carbons at $\delta = 83.6$, 88.9 and 113.0, as well as an oxymethylene carbon at $\delta = 62.5$ ppm. The signals of three methyl groups were observed at $\delta = 17.4, 17.9$ and 18.1 ppm, respectively. In the ¹H NMR spectrum three oxymethine protons appeared at $\delta = 4.05$ (dd, J = 8.0, 2.5 Hz), 4.47 (q, J = 6.0 Hz) and 5.27 ppm (1H, s). The oxymethylene protons resonated at δ = 3.60 (1H, dd, J = 12.0, 2.5 Hz) and $\delta = 3.93$ ppm (1H, dd, J = 12.0, 8.0 Hz). It further exhibited signals of three secondary methyls as doublets at δ = 0.96 (J = 7.0 Hz), 1.0 (J = 7.0 Hz) and 1.28 ppm(J = 6 Hz). The presence of a one proton septet at $\delta = 2.2$ ppm indicated the presence of an isopropyl group.

Table 1. ¹H (500 MHz) and ¹³C (125 MHz) NMR spectroscopic data of compound 1 recorded in CD₃OD.

С	δ_H (mult., J in Hz)	δ_C	HMBC (¹ H- ¹³ C)
2	4.47 (q, J = 6.0)	83.6, CH	C-2a, C-2', C-4, C-6a, C-6b
2a	-	96.5, C	_
4	_	176.2, C	_
4a	_	80.1, C	_
5	4.05 (dd, J = 2.5, 8.0)	88.9, CH	C-4, C-4a, C-5', C-6b
6a	5.27 (s)	113.0, CH	C-2, C-2a, C-4a, C-5, C-6b
6b	_	92.2, C	_
1'	1.28 (d, J = 6.0)	18.1, CH ₃	_
2'	2.16 (sept, J = 7.0)	32.2, CH	C-2, C-2a, C-3', C-4', C-6b
3′	0.96 (d, J = 7.0)	17.9, CH ₃	_
4′	1.01 (d, J = 7.0)	17.4, CH ₃	_
5′	3.60 (dd, J = 2.5, 12.0)	62.5, CH ₂	C-5
	3.93 (dd, J = 8.0,12.0)		

The assignments of the signals in the ¹H and ¹³C NMR spectra (Table 1) were made with the help of ¹H-¹H COSY, HMQC and HMBC correlations while the stereochemistry was fully authenticated by NOESY experiments (Fig. 1).

The structure and relative configuration of compound 1 are established by single-crystal X-ray diffraction studies as illustrated in Fig. 2. Although natural products with two fused furan rings have previously been reported in the literature the present work describes the first ever isolation of a new natural product with three fused furan rings [3-5].

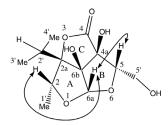


Fig. 1. Structure and important NOESY correlations of compound 1.

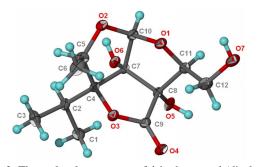


Fig. 2. The molecular structure of $\bf 1$ in the crystal (displacement ellipsoids drawn at the 30 % probability level; H atoms as spheres with arbitrary radii).

The substance was crystallized from CHCl₃/MeOH. The asymmetric unit comprised one molecule of 1 (Z=4) in the orthorhombic space group $P2_12_12_1$. Compound 1 is composed of three fused five-membered rings A (O1/C7-C8/C10-C11), B (O2/C4-C5/C7/C10) and C (O3/C4/C7/C8-C9). All three rings were found to be cis-fused to each other and adopt envelope conformations. The two hydroxy substituents were found to be pseudo-axially oriented at C-7 and C-8 with cis conformation to each other with angles of 109.5° . The hydroxymethylene moiety is attached peudo-axially to ring A whereas the isopropyl group is found to be equatorially oriented at C-4.

Experimental Section

General experimental procedures

Column chromatography (CC) was performed on silica gel (250-400 mesh; E. Merck, Darmstadt, Germany). The purity of the compounds was checked on TLC: SiO₂ 60 F₂₅₄ plates (E. Merck, Darmstadt, Germany), detection was done in iodine vapors and by spraying with ceric sulfate in 10 % H₂SO₄. Melting points were recorded in glass capillary tubes using a Büchi 535 melting point apparatus. Optical rotations were measured on a Jasco DIP-360 (Japan Spectroscopic Co. Ltd., Tokyo, Japan). The UV spectra were recorded on a Hitachi UV-3200 spectrophotometer while the IR spectrum was recorded on a Jasco 302-A spectrophotometer in KBr. The EI, FAB mass and HRMS ((+)-FAB) spectra were recorded on Jeol JMS-HX-110 and JMS-DA-500 mass spectrometers using glycerol as matrix. The ¹H and ¹³C NMR spectra were recorded on a Bruker AMX-500 MHz instrument in CD₃OD. Chemical shifts (δ) are given in ppm, and scalar coupling constant (J) are reported in Hertz.

Plant material

The plant material of *H. eichwaldi* Steud was collected in Karachi in November 2008 and identified by Prof. Dr. Surrayya Khatoon, Plant Taxonomist, Department of Botany, University of Karachi, Pakistan. A voucher specimen (no. 71489) has been deposited in the herbarium of the same Department.

Extraction and isolation

The shade-dried and chopped whole plant of H. eichwaldi (10 kg) was extracted with MeOH (3 × 25 L) at r. t. for 8 d each. The methanol extract was concentrated in vacuo. The greenish-brown residue (240 g) was divided into n-hexane-(110 g), EtOAc- (18 g), n-BuOH- (35 g), and H_2O -soluble (24 g) sub-fractions by solvent extraction. The EtOAc sub-fraction was subjected to column chromatography on silica gel using a mixture of CHCl₃-MeOH in order of in-

Table 2. Crystallographic data and numbers pertinent to data collection and structure refinement of 1.

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Formula	C ₁₂ H ₁₈ O ₇		
$M_{ m r}$	274.26		
Crystal size, mm ³	$0.40\times0.14\times0.01$		
Crystal system	orthorhombic		
Space group	$P2_12_12_1$		
a, Å	7.5102(9)		
b, Å	8.1278(10)		
c, Å	20.879(2)		
V , \mathring{A}^3	1274.5(3)		
Z	4		
$D_{\rm calcd}$, g cm ⁻³	1.43		
$\mu(\text{Mo}K_{\alpha}), \text{mm}^{-1}$	0.1		
F(000), e	584		
Wavelength; λ, Å	MoK_{α} ; 0.71073		
<i>T</i> , K	293(2)		
hkl range	$\pm 9, -9 \rightarrow +10, -25 \rightarrow +27$		
$\theta_{\rm max}$, deg	27.5		
Refl. measured / unique / R_{int}	8839 / 1707 / 0.0305		
Param. refined / restraints	183 / 16		
$R(F) / wR(F^2)^{a,b} (I \ge 2\sigma(I))$	0.0428 / 0.1082		
$R(F) / wR(F^2)^{a,b}$ (all refls.)	0.0492 / 0.1140		
$GoF(F^2)^c$	1.148		
$\Delta \rho_{\rm fin}$ (max / min), e Å ⁻³	0.32 / -0.17		

a $R1 = ||F_0| - |F_c||/\Sigma |F_0|$; b $wR2 = [\Sigma w(F_0^2 - F_c^2)^2/\Sigma w(F_0^2)^2]^{1/2}$, $w = [\sigma^2(F_0^2) + (AP)^2 + BP]^{-1}$, where $P = (\text{Max}(F_0^2, 0) + 2F_c^2)/3$; c $GoF = [\Sigma w(F_0^2 - F_c^2)^2/(n_{\text{obs}} - n_{\text{param}})]^{1/2}$.

creasing polarity to yield fractions A-D. Fraction B (9 g) which eluted with $CHCl_3$ -MeOH (9.2:0.8) was again chromatographed using a mixture of $CHCl_3$ -MeOH in increasing order of polarity to obtain fractions B_1-B_9 . Fraction B_4 (4 g) which was obtained from $CHCl_3$ -MeOH (9.2:0.8) was further chromatographed eluting with $CHCl_3$ -MeOH (9:1) to afford 1 (20 mg). Fraction B_6 (1.4 g) which was obtained with $CHCl_3$ -MeOH (8.8:1.2) was again chromatographed and eluted with $CHCl_3$ -MeOH (8.5:15) to afford the compound 2 (1.2 g). Compound 2 was identified through comparison of physical and spectral data with those reported in literature [6].

4a,6b-Dihydroxy-5-(hydroxymethyl)-2a-isopropyl-2-methyl-hexahydro-4H-1,3,6-trioxa-cyclopenta[cd]pentalen-4-one (1)

Crystalline solid; m. p. = 172 °C. $- [\alpha]_D^{25} = +18.5$ (c=0.31, MeOH). - IR (KBR): $v_{\rm max} = 3200$ (OH), 1799 (C=O), 1438, 1190, 1109, 995 cm $^{-1}$. $- ^{1}H$ NMR (500 MHz, CD₃OD) and ^{13}C NMR (125 MHz, MeOH): see Table 1. - MS (EI, 70 eV): m/z (%) = 243 (43) [M–CH₂OH] $^+$, 214 (4), 155 (40), 141 (82), 131 (78), 43 (100). - MS ((+)-FAB): m/z = 275, [M+H] $^+$. - MS ((-)-FAB): m/z = 273 [M–H] $^+$. - HRMS ((+)-FAB): m/z = 275.1130 for C₁₂H₁₉O₇, [M+H] $^+$).

Single-crystal X-ray structure determination of compound 1

Crystal data and numbers pertinent to data collection and structure refinement of $\bf 1$ are summarized in Table 2. The structure was solved by Direct Methods and expanded by Fourier techniques using SIR92 [7]. Refinement was done by full-matrix least-squares calculations on F^2 with the program SHELXL-97 [8]. In the absence of heavy atoms, the Friedel pairs were merged and no Flack refinement was undertaken. Carbon-bound hydrogen atoms were placed in calculated positions (C–H 0.93 to 0.98, O–H 0.82 Å) and were included in the refinement in the riding model approximation, with $U_{\rm iso}$ (H) set to 1.2 to 1.5 $U_{\rm eq}$ (C,O).

Two ether oxygen atoms were disordered over two positions in a ratio of 0.625(5)/0.375. Their anisotropic displacement factors were restrained to be nearly isotropic and set to be similar. Fig. 2 shows the molecular structure.

CCDC 843754 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre *via* www.ccdc.cam.ac.uk/data_request/cif.

Acknowledgements

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