Tuning the Photophysical Properties of Cyclometalated Ir(III) Complexes by a Trifluoroacetyl Group

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Four cationic Ir(III) complexes, $[Ir(dpq)_2(bpy)]PF_6$ (1), $[Ir(dpq)_2(phen)]PF_6$ (2), $[Ir(tfapq)_2(bpy)]PF_6$ (3), and $[Ir(tfapq)_2(phen)]PF_6$ (4) (dpqH=2,4-diphenylquinoline, tfapqH=2-(4'-trifluoroacetylphenyl)-4-phenylquinoline, bpy = 2,2'-bipyridine, phen = 1,10-phenanthroline) have been synthesized and fully characterized. The structure of 4 was also confirmed by single-crystal X-ray diffraction. The electron-acceptor character of the trifluoroacetyl unit leads to a reduced HOMO-LUMO gap and consequently a red-shift of the UV/Vis absorption and luminescence spectra. The solvophobic

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character of the trifluoroacetyl unit gives rise to a molecule assembly in solution.

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