Influence of Anions on the Self-assembly of Cd(II)-containing Coordination Polymers Based on the Flexible Ligand 2-((1*H*-1,2,4-Triazol-1-yl)methyl)-1*H*-benzimidazole

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Three Cd(II)-containing coordination polymers $[CdCl_2(tmb)]_n$ (1) and two polymorphs of $\{[CdI_2(tmb)]\cdot DMF\}_n$ (2, 3) $(tmb = 2\cdot((1H-1,2,4-triazol-1-yl)methyl)\cdot 1H$ -benzimidazole) have been synthesized by the reactions of tmb with the cadmium salts. Polymer 1 exhibits an infinite \cdots Cd- $(Cl1)_2$ -Cd- $(Cl2)_2$ -Cd- \cdots chain with μ_2 Cl bridges, while polymers 2 and 3 are isomers, in which the Cd(II) ions are bridged by the bidentate tmb ligands leading to \cdots Cd-tmb-Cd- \cdots chains. The polymers form 3-D supramolecular frameworks through hydrogen bonds and $\pi \cdots \pi$ stacking interactions. The different structures of polymers 1-3 indicate that the anions and the flexibility of the tmb ligand can influence the structures of the coordination polymers. The infrared spectra and luminescent

properties of the polymers have been investigated in the solid state at room temperature. *Key words:* Crystal Structure, Anion Influence, Flexibility of Ligand, Luminescence