# Syntheses, Spectroscopy and Crystal Structures of ( $R$ )- N -(1-Aryl-ethyl)salicylaldimines and $\left[\operatorname{Rh}\{(R)-N\right.$-(1-aryl-ethyl)salicylaldiminato $\}\left(\boldsymbol{\eta}^{4}\right.$-cod)] Complexes 

Mohammed Enamullah ${ }^{\text {a }}$, A. K. M. Royhan Uddin ${ }^{\text {a }}$, Anne-Christine Chamayou ${ }^{\text {b }}$, and Christoph Janiak ${ }^{\text {b }}$<br>a Department of Chemistry, Jahangirnagar University, Dhaka-1342, Bangladesh<br>${ }^{\text {b }}$ Institut für Anorganische und Analytische Chemie, Universität Freiburg, Albertstr. 21, D-79104 Freiburg, Germany

Reprint requests to Prof. M. Enamullah. Fax: +8802-7708069.
E-mail: menam@juniv.edu/enamullahju@yahoo.com or to
Prof. C. Janiak. Fax: +49-7612036147. E-mail: janiak @ uni-freiburg.de
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Condensation of salicylaldehyde with enantiopure ( $R$ )-(1-aryl-ethyl)amines yields the enantiopure Schiff bases $(R)-N$-(1-aryl-ethyl)salicylaldimine (HSB*; aryl = phenyl, 2-methoxyphenyl, 3methoxyphenyl, 4-methoxyphenyl (4), 4-bromophenyl (5), 2-naphthyl). These Schiff bases readily react with dinuclear (acetato) $\left(\eta^{4}\right.$-cycloocta-1,5-diene)rhodium $(\mathrm{I}),\left[\operatorname{Rh}\left(\mu-\mathrm{O}_{2} \mathrm{CMe}\right)\left(\eta^{4} \text {-cod) }\right]_{2}\right.$, to afford the mononuclear complexes, cyclooctadiene-((R)-N-(1-aryl-ethyl)salicylaldiminato- $\left.\kappa^{2} N, O\right)$ rhodium $(\mathrm{I}),\left[\mathrm{Rh}\left(\mathrm{SB}^{*}\right)\left(\eta^{4}-\mathrm{cod}\right)\right]\left(\mathrm{SB}^{*}=\right.$ deprotonated chiral Schiff base $=$ salicylaldiminate; aryl $=$ phenyl (7), 2-methoxyphenyl, 4-methoxyphenyl, 4-bromophenyl, 2-naphthyl). The complexes have been characterized by IR, UV/vis, ${ }^{1} \mathrm{H} /{ }^{13} \mathrm{C}$ NMR and mass spectrometry, optical rotation as well as by single-crystal X-ray structure determination for $\mathbf{4}, \mathbf{5}$ and 7 . The structure of $\mathbf{5}$ shows $\mathrm{C}-\mathrm{Br} \cdots \pi$ contacts. Compound 7 is only the second example of a $\operatorname{Rh}\left(\eta^{4}\right.$-cod) complex with a six-membered $\mathrm{Rh}-\mathrm{N}, \mathrm{O}$-chelate ring.

Key words: $(R)$-Schiff Bases, $\operatorname{Rh}\left(\eta^{4}\right.$-cod) Complexes, Chelate Complexes, $\pi$ Interactions, Optical Activity, Chirality

## Introduction

The synthesis of chiral metal complexes is of constant interest [1]. There are continuous developments of optically active Schiff base ligands (HSB*) and their transition metal complexes for applications as chiral catalysts [2-9]. Examples of organometallic compounds with HSB* ligands are the half-sandwich complexes $\left[\operatorname{Ru}\left(\mathrm{SB}^{*}\right) X\left(\eta^{6}\right.\right.$-benzene $\left.)\right]\left\{\mathrm{SB}^{*}=(S)-N\right.$ -1-phenylethylsalicylaldiminate; $X=\mathrm{Cl}, 4-/ 2-\mathrm{Me}-\mathrm{py}$, $\left.\mathrm{PPh}_{3}\right\},\left[\mathrm{M}\left(\mathrm{SB}^{*}\right) X\left(\eta^{6}\right.\right.$-arene $\left.)\right](\mathrm{M}=\mathrm{Ru}(\mathrm{II}), \mathrm{Os}(\mathrm{II})$; $X=\mathrm{Cl}, \mathrm{I})[10,11],\left[\mathrm{Ru}\left(\mathrm{SB}^{*}\right) X\left(\eta^{6}-p\right.\right.$-cymene $\left.)\right](X=$ various monodentate ligands) $[12,13]$, and $\left[\mathrm{Rh}\left(\mathrm{SB}^{*}\right)\right.$ -$\left(\eta^{4}\right.$-cod $\left.)\right]\left\{\mathrm{SB}^{*}=(S)-(\alpha)\right.$-(2-pyridyl)-salicylaldiminate\} [14].

Bidentate (HSB) and tetradentate $\left(\mathrm{H}_{2} \mathrm{SB}\right)$ Schiff bases react easily with dinuclear $\left[\operatorname{Rh}(\mu-X)\left(\eta^{4} \text {-cod }\right)\right]_{2}$ ( $X=\mathrm{Cl}, \mathrm{OMe}, \mathrm{O}_{2} \mathrm{CMe}$; cod $=1,5$-cyclooctadiene) to give mononuclear $\left[\operatorname{Rh}(\mathrm{SB})\left(\eta^{4}\right.\right.$-cod) $)$ ( $\mathrm{SB}=$ salicylaldiminate) and dinuclear $\left[\left\{\operatorname{Rh}\left(\eta^{4}-\operatorname{cod}\right)\right\}_{2}(\mathrm{SB})\right](\mathrm{SB}=$
bis-salicylaldiminate) complexes [14-20]. We recently synthesized $\operatorname{Rh}\left(\eta^{4}\right.$-cod) complexes containing chiral amino acids, chiral amino alcohols and tetradentate Schiff bases as co-ligands starting from dinuclear $\left[\mathrm{Rh}\left(\mu-\mathrm{O}_{2} \mathrm{CMe}\right)\left(\eta^{4} \text {-cod }\right)\right]_{2}[21-24]$. In continuation, we report here the syntheses and characterizations of enantiopure Schiff base compounds HSB* and their $\left[\operatorname{Rh}\left(\mathrm{SB}^{*}\right)\left(\eta^{4}\right.\right.$-cod $\left.)\right]$ complexes $\left[\mathrm{SB}^{*}=(R)\right.$ -$N$-(1-aryl-ethyl)salicylaldiminate, with $X=$ phenyl, 2methoxyphenyl, 3-methoxyphenyl, 4-methoxyphenyl, 4-bromophenyl, 2-naphthyl].

## Results and Discussion

Condensation of the salicylaldehyde with enantiopure $(R)-(1$-aryl-ethyl)amines yields the optically active ( $R$ )- $N$-(1-aryl-ethyl)salicylaldimines [HSB*; aryl $=$ phenyl (1), 2-methoxyphenyl (2), 3-methoxyphenyl (3), 4-methoxyphenyl (4), 4-bromophenyl (5), 2-naphthyl (6)] (Scheme 1). Reaction of dinuclear


Scheme 1. Synthetic route to ( $R$ )- $N$-(1-aryl-ethyl)salicylaldimines (HSB*; 1-6).

Scheme 2. Synthetic route to $\operatorname{Rh}\left(\mathrm{SB}^{*}\right)\left(\eta^{4}-\operatorname{cod}\right)(7-11)$.
$\left[\operatorname{Rh}\left(\mu-\mathrm{O}_{2} \mathrm{CMe}\right)\left(\eta^{4} \text {-cod }\right)\right]_{2} \quad(\operatorname{cod}=1,5-$ cyclooctadiene) with ( $R$ )- $N$-(1-aryl-ethyl)salicylaldimine in toluene $/ \mathrm{MeOH}$ affords the mononuclear complexes, cyclooctadiene- $\{(R)$ - $N$-(1-aryl-ethyl)salicylaldimin-ato- $\left.\kappa^{2} N, O\right\}$-rhodium $(\mathrm{I}),\left[\operatorname{Rh}\left(\mathrm{SB}^{*}\right)\left(\eta^{4}\right.\right.$-cod $\left.)\right]\left(\mathrm{SB}^{*}=\right.$ deprotonated chiral Schiff base $=$ salicylaldiminate) (7-11), in Scheme 2.

The ${ }^{1} \mathrm{H} /{ }^{13} \mathrm{C}$ NMR spectra of the Schiff bases $\mathbf{1 - 6}$ and their complexes $\mathbf{7 - 1 1}$ correspond well to those of related compounds [2,3,9-11, 25-35]. The presence of $o / m / p-\mathrm{OCH}_{3}, p-\mathrm{Br}$ and 2-naphthyl groups in 2-6 shifts the proton signals downfield by $0.1-0.5 \mathrm{ppm}$ in contrast to those in $\mathbf{1}$ due to their electron donating inductive effect.

In the ${ }^{1} \mathrm{H}$ NMR spectra the signals for the exo- and endo-methylene protons of the rhodium-coordinated 1,5 -cyclooctadiene in $\left[\operatorname{Rh}\left(\mathrm{SB}^{*}\right)\left(\eta^{4}\right.\right.$-cod $\left.)\right](7-11)$ each appear as multiplets at about 1.90 and 2.40 ppm , respectively. The olefinic protons show two multiplets at 3.6-3.7 and $4.5-4.6 \mathrm{ppm}$ (except for 8, see be-
low). The upfield resonance at $3.6-3.7 \mathrm{ppm}$ is assigned to protons 'trans to O ', and the downfield resonance at $4.5-4.6 \mathrm{ppm}$ to protons 'trans to N ' [20, $24-$ 28, 30-35].

Complex $\left[\operatorname{Rh}\left(\mathrm{SB}^{*}\right)\left(\eta^{4}\right.\right.$-cod)] (8) with $\mathrm{SB}^{*}=$ ( $R$ )- N -(1-(2-methoxyphenyl-ethyl)salicylaldiminate shows four multiplets at $3.7,4.3,4.4$ and 4.5 ppm . The ortho $-\mathrm{OCH}_{3}$ substituent on the phenyl ligand leads to stronger steric interactions with the olefin protons in comparison to metalpara- $\mathrm{OCH}_{3}$ and thereby creates sufficient differences in chemical shifts between 'left' and 'right' protons. Similar olefin proton resonances are observed in $[\mathrm{M}(\mathrm{sal}=\mathrm{N}$ $o / p$-toluene $)\left(\eta^{4}\right.$-cod)] ( $\mathrm{M}=\mathrm{Rh}$, Ir) [20], showing three multiplets for $o$-toluene and two multiplets for $p$-toluene (see Table 1). Also, the dinuclear complexes $\left[\left(\operatorname{Rh}\left(\eta^{4} \text {-cod }\right)\right)_{2}(\right.$ salophen $\left.)\right][24]$ and $[\operatorname{Rh}(\mu-\mathrm{hp} / \mu$-mhp) ( $\eta^{4}$-cod) $]_{2}$ [27] show four multiplets.

In $\mathrm{CDCl}_{3}$ the proton signal for $\mathrm{CH}=\mathrm{N}$ of the Schiff bases at $8.2-8.4 \mathrm{ppm}$ is shifted upon Rh complexa-

Table 1. ${ }^{13} \mathrm{C}$ NMR spectral data ( $\delta$ in ppm) and $J\left({ }^{103} \mathrm{Rh}-{ }^{13} \mathrm{C}\right)(\mathrm{Hz})$ in the cod region in $\mathrm{Rh}\left(\eta^{4}\right.$-cod) complexes in $\mathrm{CDCl}_{3}$ (unless noted otherwise).

| complexes | methylene carbons | olefinic carbons $\left(J\left({ }^{103} \mathrm{Rh}-{ }^{13} \mathrm{C}\right)\right.$ in parentheses) |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  |  | trans to N |  | trans to O |  |
|  |  | 'left' | 'right' ${ }^{\text {a }}$ | 'left' | 'right' ${ }^{\text {a }}$ |
| $\overline{\left[\operatorname{Rh}\left(\mathrm{SB}^{*}\right)\left(\eta^{4}-\mathrm{cod}\right)\right](7)}$ | 32.5, 32.0, 29.6, 29.2 | 85.7 (12.1) | 85.3 (12.3) | 73.5 (14.2) | 71.4 (14.6) |
| $\left[\operatorname{Rh}\left(\mathrm{SB}^{*}\right)\left(\eta^{4}\right.\right.$-cod) $](\mathbf{8})$ | 33.1, 31.7, 30.1, 28.9 | 85.0 (12.6) | 84.0 (12.0) | 74.7 (13.4) | 71.6 (14.5) |
| $\left[\mathrm{Rh}\left(\mathrm{SB}^{*}\right)\left(\eta^{4}\right.\right.$-cod) $](9)$ | 32.5, 32.0, 29.6, 29.1 | 85.7 (12.2) | 85.3 (12.2) | 73.6 (14.0) | 71.2 (14.6) |
| $\left[\operatorname{Rh}\left(\mathrm{SB}^{*}\right)\left(\eta^{4}\right.\right.$-cod) $](\mathbf{1 0})$ | 31.1, 30.7, 28.2, 27.9 | 84.5 (11.8) | 84.2 (12.2) | 72.0 (14.6) | 70.2 (14.1) |
| $\left[\operatorname{Rh}\left(\mathrm{SB}^{*}\right)\left(\eta^{4}\right.\right.$-cod) $](11)$ | 32.6, 32.1, 29.6, 29.2 | 85.8 (11.6) | 85.4 (12.3) | 73.7 (14.3) | 71.4 (14.2) |
| $\left[\mathrm{Rh}(\mathrm{N}, \mathrm{O})\left(\eta^{4}\right.\right.$-cod) ${ }^{\text {[ } 35]}{ }^{\text {b }}$ | 32.1, 31.9, 29.6, 29.5 | 81.6 | 81.3 | 75.4 | 75.1 |
| [ $\mathrm{Rh}($ sal $=\mathrm{N}-o-$ tol $)\left(\eta^{4}\right.$-cod) $]$ [20] | 31.7, 31.3, 29.3, 28.8 | 85.1 (12.5) | 84.6 (12.5) | 74 (17.5) | 72.5 (15.0) |
| [(Rh( $\eta^{4}$-cod) $)_{2}$ (salophen)] [24] | 32.6, 30.3, 29.5, 27.9 | 85.8 (11.7) | 84.3 (11.8) | 74.3 (14.6) | 69.7 (14.4) |
| [(Rh( $\eta^{4}$-cod) $)_{2}$ (salophen)] [20] | 32.5, 30.3, 29.5, 27.9 | 85.8 (12.5) | 84.3 (12.5) | 74.3 (15.0) | 69.7 (15.0) |
| $\left[\operatorname{Rh}(\mu-\mathrm{hp} /-\mathrm{mhp})\left(\eta^{4} \text {-cod }\right)\right]_{2}[27]^{\text {c }}$ | $\begin{aligned} & 35.0,33.0,30.1,29.0 / 33.4 \text {, } \\ & 32.1,30.5,29.2 \end{aligned}$ | 89.1/87.7 | 772/76.6 | 74.4/72.8 | 70.9/72.2 |
| $\left[\mathrm{Rh}\left(\mathrm{sal}=\mathrm{N}-\mathrm{CH}_{3} /-\mathrm{Ph}\right)\left(\eta^{4}\right.\right.$-cod) $][20]$ | 32.1, 28.9/31.3, 29.0 | 85.3 (12.5) | 4.7 (12.5) | 72.8 (12 | 3.0 (12.5) |
| [ $\mathrm{Rh}(\mathrm{sal}=\mathrm{N}-p-\mathrm{tol})\left(\eta^{4}\right.$-cod)] [20] | 31.4, 29.0 |  | 2.5) |  | (15) |
| [ $\mathrm{Rh}\left(o-\mathrm{O}_{2} \mathrm{NC}_{6} \mathrm{H}_{4} \mathrm{NH}\right)\left(\eta^{4}\right.$-cod) $]$ [28] | 31.3, 29.4 |  |  |  |  |
| [ $\left\{\mathrm{Rh}\left(\eta^{4} \text {-cod) }\right\}_{2}(\mathrm{dcbi})\right]\left(\mathrm{NHEt}_{3}\right)$ [29] | 31.2, 30.0 |  | 13) |  |  |
| $\underline{[ } 2 \mathrm{Rh}\left(\eta^{4}\right.$-cod $\left.)\right\}_{2}$ (salen) $][24]$ | 31.7, 28.8 |  | 1.9) |  | 14.2) |

${ }^{\mathrm{a}}$ 'left' and 'right' is an arbitrary assignment for the olefinic carbons to either side of a plane bisecting the $\mathrm{C}=\mathrm{C}$ bond; ${ }^{\mathrm{b}}$ in $\mathrm{C}_{6} \mathrm{D}_{6}$; ${ }^{\mathrm{c}}$ in [ $\mathrm{D}_{8}$ ]toluene.
tion to higher field (at 7.8 ppm ) and splits into a symmetrical doublet by about $2.0 \mathrm{~Hz}(J)$ due to ${ }^{103} \mathrm{Rh}-{ }^{1} \mathrm{H}$ coupling [17, 19]. In [D $\mathrm{D}_{6}$ ]DMSO this signal remains a singlet at $8.0-8.1 \mathrm{ppm}$ for the complexes.

In the ${ }^{13} \mathrm{C}$ NMR spectra the cod methylene carbon atoms in 7-11 give four singlets of equal intensity at $\delta=29-31 \mathrm{ppm}$ in contrast to only one singlet in $\left[\mathrm{Rh}\left(\mathrm{O}_{2} \mathrm{CMe}\right)\left(\eta^{4} \text {-cod }\right)\right]_{2}$ [22] and $[\mathrm{Rh}($ aminocarboxylato) ( $\eta^{4}$-cod)] [22, 23]. Similarly, the four olefinic carbon atoms of cod give four doublets due to ${ }^{103} \mathrm{Rh}-$ ${ }^{13} \mathrm{C}$ coupling, two at lower field ( $84-86 \mathrm{ppm}$ ) which are assigned to ' C trans to N ', the other two at higher field ( $70-75 \mathrm{ppm}$ ) assigned to 'C trans to O ' (see Table 1) [16, 20, 24, 26, 29, 30]. The observed ${ }^{103} \mathrm{Rh}-$ ${ }^{13} \mathrm{C}$ (olefin) spin-spin coupling constants for ' C trans to N ' $(c a . J=12 \mathrm{~Hz})$ and 'C trans to O ' ( $c a . J=$ 14 Hz ) agree with data for related mononuclear $\operatorname{Rh}\left(\eta^{4}\right.$ cod) complexes [16, 20, 24, 27, 30, 35] (see Table 1). The occurrence of four singlets and four doublets is explained by steric and magnetic anisotropy effects in addition to the trans influence of the coordinated $\mathrm{N}, \mathrm{O}$ chelate on the carbon resonances [27]. The observed chemical shift difference between the 'left' and 'right' carbon atoms trans to the same donor atom are larger for 'trans to O ' than for 'trans to N ' in 7-11.

Mass spectra of the Schiff bases $\mathbf{1 - 6}$ and the $\left[\operatorname{Rh}\left(\mathrm{SB}^{*}\right)\left(\eta^{4}\right.\right.$-cod $\left.)\right]$ complexes $\mathbf{7 - 1 1}$ show the parent ion peaks. UV/vis Electronic spectra of the rhodium complexes feature two broad bands with absorption
maxima at $\lambda_{\max }=234-244 \mathrm{~nm}\left(\varepsilon_{\max }=23750-\right.$ $59700 \mathrm{~L} \cdot \mathrm{~mol}^{-1} \cdot \mathrm{~cm}^{-1}$ ), associated with the intra-ligand $\pi \rightarrow \pi^{*}$ transition, and at $\lambda_{\text {max }}=388-394 \mathrm{~nm}$ $\left(\varepsilon_{\text {max }}=5000-14700 \mathrm{~L} \cdot \mathrm{~mol}^{-1} \cdot \mathrm{~cm}^{-1}\right)$, associated with the metal-to-ligand charge transfer (MLCT) transitions of $\mathrm{Rh} \rightarrow\left(\eta^{4}\right.$-cod $)$ and $\mathrm{Rh} \rightarrow \mathrm{SB}^{*}$ [21-23]. The polarimetric measurements in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ or $\mathrm{CH}_{3} \mathrm{Cl}$ exhibit rotations to the left between $-95^{\circ}$ and $-170^{\circ}$ at 578 nm and $20^{\circ} \mathrm{C}$ for enantiopure $R$-Schiff bases, and rotations to the right between $+200^{\circ}$ and $+333^{\circ}$ at 578 nm and $20^{\circ} \mathrm{C}$ for the $\mathrm{Rh}\left(R-\mathrm{SB}^{*}\right)$-complexes.

The single-crystal structures of the enantiopure Schiff bases 4 and 5 confirm the molecular composition and absolute configuration. The molecular structures are depicted in Figs. 1 and 2, respectively. Bond


Fig. 1. Molecular structure of $\mathbf{4}$ with intramolecular hydrogen bond. Thermal ellipsoids with $50 \%$ probability. Selected bond lengths (A) and angles (deg): C13-O2 1.381(3), C7-N 1.275(3), N-C8 1.481(3); C7-N-C8 119.1(2). Hydrogen bonding interaction (dashed line) as $\mathrm{O}-\mathrm{H}, \mathrm{H} \cdots \mathrm{N}$, $\mathrm{O} \cdots \mathrm{N}, \mathrm{O}-\mathrm{H} \cdots \mathrm{N}\left(\AA,{ }^{\circ}\right): 1.00(3), 1.68(5), 2.580(2), 148(3)$.


Scheme 3. Bond lengths ( $\AA$ ) for $\mathrm{Rh}-\mathrm{C}_{\mathrm{cod}}$ and $\mathrm{C}=\mathrm{C}_{\text {cod }}$ in the two symmetry-independent molecules in 7.


Fig. 2. Molecular structure of 5 with intramolecular hydrogen bond. Thermal ellipsoids with $50 \%$ probability. Selected bond lengths $(\AA)$ and angles (deg): C13-Br 1.908(3), C7-N $1.275(3), \mathrm{N}-\mathrm{C} 8$ 1.473(3); $\mathrm{C} 7-\mathrm{N}-\mathrm{C} 8$ 118.5(2). Hydrogen bonding interaction (dashed line) as $\mathrm{O}-\mathrm{H}, \mathrm{H} \cdots \mathrm{N}, \mathrm{O} \cdots \mathrm{N}$, $\mathrm{O}-\mathrm{H} \cdots \mathrm{N}(\AA, \operatorname{deg}): 0.92(4), 1.72(4), 2.590(3), 156(4)$.
lengths are within the expected range. The expected intramolecular hydrogen bond is observed between the salicyl-OH group and the imine nitrogen atom [36].

The molecular packing of $\mathbf{4}$ does not show $\pi-\pi$ interactions [37-39] but only a C-H $\cdots \pi$ interaction $\mathrm{C} 16-\mathrm{H} \cdots(\mathrm{C} 10-\mathrm{C} 15)$ with $\mathrm{H} \cdots$ centroid $2.95 \AA$ and $\mathrm{C}-$ $\mathrm{H} \cdots \pi$ plane $60^{\circ}$ [39-42]. The molecular packing in the structure of 5 is influenced by a C-H $\cdots \pi$ interaction C12-H $\cdots(\mathrm{C} 1-\mathrm{C} 6)$ with $\mathrm{H} \cdots$ centroid $2.71 \AA$, C$\mathrm{H} \cdots$ centroid $138^{\circ}$ and $\mathrm{C}-\mathrm{H} \cdots \pi$ plane $52^{\circ}$, and also by $\mathrm{C}-\mathrm{Br} \cdots \pi$ contacts to the salicyl ring (C1-C6) with $\mathrm{Br} \cdots$ centroid $3.816(1) \AA, \mathrm{C}-\mathrm{Br} \cdots$ centroid $166.0^{\circ}$ and C-Br $\cdots \pi$ plane $73.4^{\circ}$ as illustrated in Fig. 3 [43].

The molecular structure of the rhodium complex 7 proves the suggested $N, O$-chelation of the deprotonated Schiff base salicylaldiminato ligand (Fig. 4). Again bond lengths and their variations are as expected [ $12,16,23,24,26,27]$. Compound 7 is only the second example of a $\operatorname{Rh}\left(\eta^{4}\right.$-cod) complex with a six-membered $\mathrm{Rh}-\mathrm{N}, \mathrm{O}$-chelate ring. The other example is the dinuclear compound $\left[\left\{\operatorname{Rh}\left(\eta^{4}-\operatorname{cod}\right)\right\}_{2}\left(N, N^{\prime}\right.\right.$ ( 1,2 -phenylene)bis-(salicylaldiminato))] with an achiral tetradentate Schiff base ligand [15, 16, 44]. The cod-ligand in 7 is bound slightly asymmetrically (Scheme 3) which reflects the different trans nitrogen


Fig. 3. Packing diagram of 5 to illustrate the $\mathrm{C}-\mathrm{H} \cdots \pi$ and $\mathrm{C}-$ $\mathrm{Br} \cdots \pi$ contacts as dashed lines to the salicyl ring centroid.


Fig. 4. Molecular structure of the two symmetry-related molecules of 7. Selected bond lengths ( A ) and angles (deg): Rh1-O1 2.0268(13), Rh1-N1 2.085(2), Rh1-C cod $^{2.118(3)-~}$ 2.161(3), Rh2-O2 2.0388(13), Rh2-N2 2.0840(19), Rh2$\mathrm{C}_{\mathrm{cod}} 2.117(4)-2.168(4) ; \mathrm{O} 1-\mathrm{Rh} 1-\mathrm{N} 1$ 90.93(7), O2-Rh2-N2 90.28(6).
or oxygen donor atoms and the 'left' and 'right' differentiation as mirrored in the four olefinic ${ }^{13} \mathrm{C}$ NMR resonances.
The unit cell in the crystal structure of $\mathbf{7}$ contains two symmetry-independent molecules which superficially appear related by a pseudo two-fold axis. No classical hydrogen bonds, $\pi-\pi$ interactions or $\mathrm{C}-\mathrm{H} \cdots \pi$
contacts are discernible in 7. Van der Waals interactions between the molecules of 7 with their hydrophobic surface seem to control the packing.

## Experimental Section

All reactions were carried out under an atmosphere of dry nitrogen using standard Schlenk techniques. Solvents were dried and distilled under nitrogen prior to use: toluene, diethyl ether over Na metal; methanol over CaO ; chloroform over $\mathrm{CaCl}_{2}$. IR spectra were recorded on a Bruker Optik IFS 25 spectrometer from KBr disks at ambient temperature. UV/vis Spectra were obtained with a Shimadzu UV 3150 spectrophotometer in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ at $20^{\circ} \mathrm{C}$. Elemental analyses were carried out on a Vario EL instrument from Elementaranalysensysteme GmbH. NMR Spectra were run on a Bruker Avance DPX 200 spectrometer operating at 200 MHz $\left({ }^{1} \mathrm{H}\right)$ and $50 \mathrm{MHz}\left({ }^{13} \mathrm{C}\right)$ at $25{ }^{\circ} \mathrm{C}$ with calibration against the residual protonated solvent signal $\left(\mathrm{CDCl}_{3}: 7.26\left({ }^{1} \mathrm{H}\right)\right.$ and $77.0\left({ }^{13} \mathrm{C}\right)$; $\left[\mathrm{D}_{6}\right]$ DMSO: $2.52\left({ }^{1} \mathrm{H}\right)$ and $\left.39.5\left({ }^{13} \mathrm{C}\right) \mathrm{ppm}\right)$. The NMR grade solvents $\mathrm{CDCl}_{3}$ and $\left[\mathrm{D}_{6}\right]$ DMSO were deoxygenated prior to use. EI- and CI-MS: Thermo-Finnigan TSQ 700, with $\mathrm{NH}_{3}$ as ionization gas for CI. Polarimetric measurements were carried with a Perkin-Elmer 241 instrument in $\mathrm{CHCl}_{3}$ and $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ at $20^{\circ} \mathrm{C}$, and the values of $[\alpha]^{20}$ were determined according to the literature [10]. The starting dinuclear $\left[\mathrm{Rh}\left(\mathrm{O}_{2} \mathrm{CMe}\right)\left(\eta^{4} \text {-cod }\right)\right]_{2}$ complex was synthesized from $\left[\mathrm{RhCl}\left(\eta^{4} \text {-cod }\right)\right]_{2}[45]$ according to the literature [22, 46]. The enantiopure amines $(R)$-1-phenyl-ethylamine, $(R)$-( 2 -methoxyphenyl)ethylamine, $(R)$-( 3 -methoxyphenyl)ethylamine, $(R)$-(4-methoxyphenyl)ethylamine, $(R)$ -(4-bromophenyl)ethylamine and ( $R$ )-(2-naphthyl)ethylamine were used as received from BASF, Ludwigshafen, Germany.

## ( $R$ )-N-(1-Phenylethyl)salicylaldimine (1)

Salicylaldehyde ( $8.35 \mathrm{~mL}, 78.36 \mathrm{mmol}$ ) was dissolved into 20 mL of methanol with $2-3$ drops of conc. $\mathrm{H}_{2} \mathrm{SO}_{4}$ added into the solution which was then stirred for 10 min at r.t. An equimolar amount of $(R)$-1-phenyl-ethylamine $(10 \mathrm{~mL}, 78.39 \mathrm{mmol})$ was added to the solution. The colour soon changed to bright yellow, and the mixture was refluxed for $5-6 \mathrm{~h}$. Then, the solvent was evaporated to a volume of to $50 \%$ in vacuo and the yellow solution was left standing at r. t. for crystallization through slow solvent evaporation. After 2-3d, bright-yellow crystals suitable for X-ray measurements were obtained. The crystals were washed three times with $\mathrm{MeOH}\left(5 \mathrm{~mL}\right.$ each) and dried in vacuo at $40-50{ }^{\circ} \mathrm{C}$ for $5-6 \mathrm{~h}$ to give a bright-yellow product. Yield: 16.60 g ( $94 \%$ ) (based on salicylaldehyde). $-[\alpha]^{20}\left(c=0.84, \mathrm{CHCl}_{3}\right)$ : $-95^{\circ}(578 \mathrm{~nm}) .-\mathrm{IR}(\mathrm{KBr}): v=3063 \mathrm{~m}, 3034 \mathrm{~m}(\mathrm{H}-\mathrm{Ar})$, 1627 vs (C=N), $1578(\mathrm{C}=\mathrm{C}) \mathrm{cm}^{-1}$. - ${ }^{1} \mathrm{H}$ NMR ( 200 MHz , [D $\mathrm{D}_{6}$ ]DMSO): $\delta=1.59$ (d, $J_{\mathrm{HH}}=6.7 / 6.8^{\mathrm{a}} \mathrm{Hz}, 3 \mathrm{H}, \mathrm{H} 9$ ), 4.70 $\left(\mathrm{q}, J_{\mathrm{HH}}=6.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 8\right), 6.93\left(\mathrm{dd}, J_{\mathrm{HH}}=7.7,7.2 \mathrm{~Hz}, J_{\mathrm{HH}}=\right.$

$1.1 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{H} 4,6-\mathrm{sal}), 7.31-7.51$ (m, 7H, sal+Ph), 8.70 (s, $1 \mathrm{H}, \mathrm{H} 7$ ), 13.55 (br, $1 \mathrm{H}, \mathrm{OH}$ ). - ${ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=1.46\left(\mathrm{~d}, J_{\mathrm{HH}}=6.7 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H} 9\right), 4.37\left(\mathrm{q}, J_{\mathrm{HH}}=6.7 \mathrm{~Hz}\right.$, $1 \mathrm{H}, \mathrm{H} 8), 6.69$ (ddd, $J_{\mathrm{HH}}=7.7,7.4 \mathrm{~Hz}, J_{\mathrm{HH}}=1.0 \mathrm{~Hz}, 1 \mathrm{H}$, $\mathrm{H} 4), 6.79\left(\mathrm{~d}, J_{\mathrm{HH}}=7.9 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 6\right), 7.03-7.22(\mathrm{~m}, 7 \mathrm{H}$, $\mathrm{sal}+\mathrm{Ph}$ ), 8.22 (s, $1 \mathrm{H}, \mathrm{H} 7$ ), 13.43 (br, $1 \mathrm{H}, \mathrm{OH}$ ). - ${ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=24.9$ (C9), 68.5 (C8), $117.0(\mathrm{C} 3)$, 118.6 (C5), 118.9 (C1), 126.4 (C11,15), 127.3 (C13), 128.7 (C12,14), 131.4 (C6), 132.3 (C4), 143.9 (C10), 161.1 (C2), 163.5 (C7). - MS (EI, 70 eV ): $\mathrm{m} / \mathrm{z}(\%)=225(100)[\mathrm{M}]^{+}$, 121 (65) $\left[\mathrm{M}-\mathrm{CH}_{3} \mathrm{CC}_{6} \mathrm{H}_{5}\right]^{+}, 105$ (100) $\left[\mathrm{CH}_{3} \mathrm{CHC}_{6} \mathrm{H}_{5}\right]^{+}, 77$ (10) $\left[\mathrm{C}_{6} \mathrm{H}_{5}\right]^{+} .-\mathrm{C}_{15} \mathrm{H}_{15} \mathrm{NO}$ (225.29): calcd. C 79.97, H 6.71, N 6.22; found C 79.15, H 6.91, N 6.44 .

Compounds 2-6 were prepared following the same procedure as described for $\mathbf{1}$ using ( $R$ )-1-(2-methoxyphenyl) ethylamine, $(R)$-1-(3-methoxyphenyl)ethylamine, $(R)$-1-(4methoxyphenyl)ethylamine, ( $R$ )-1-(4-bromophenyl)ethylamine, and ( $R$ )-1-(2-naphthyl)ethylamine, respectively.
( $R$ )-N-(1-(2-Methoxyphenyl)ethyl)salicylaldimine (2)
Yield: $18.0 \mathrm{~g}(90 \%) .-[\alpha]^{20}\left(c=0.49, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right):-163^{\circ}$ $(578 \mathrm{~nm}),-255^{\circ}(546 \mathrm{~nm})$. $-\mathrm{IR}(\mathrm{KBr}): v=3054 \mathrm{~m}(\mathrm{H}-\mathrm{Ar})$, 1626 vs $(\mathrm{C}=\mathrm{N}), 1578$ vs $(\mathrm{C}=\mathrm{C}) \mathrm{cm}^{-1} .-{ }^{1} \mathrm{H}$ NMR ( 200 MHz , $\mathrm{CDCl}_{3}$ ): $\delta=1.65$ (d, $\left.J_{\mathrm{HH}}=6.6 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H} 9\right), 3.91(\mathrm{~s}, 3 \mathrm{H}$, $\mathrm{H} 16), 5.05\left(\mathrm{q}, J_{\mathrm{HH}}=6.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 8\right), 6.92\left(\mathrm{ddd}, J_{\mathrm{HH}}=\right.$ $\left.8.4,7.6 \mathrm{~Hz}, J_{\mathrm{HH}}=1.0 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{H} 4,13\right), 6.98\left(\mathrm{~d}, J_{\mathrm{HH}}=\right.$ $6.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 6), 7.05\left(\mathrm{dd}, J_{\mathrm{HH}}=6.5 \mathrm{~Hz}, J_{\mathrm{HH}}=1.4 \mathrm{~Hz}\right.$, $1 \mathrm{H}, \mathrm{H} 3), 7.29$ (d, $J_{\mathrm{HH}}=7.6 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{H} 12,14$ ), 7.37 (ddd, $\left.J_{\mathrm{HH}}=8.0,7.5 \mathrm{~Hz}, J_{\mathrm{HH}}=1.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5\right), 7.49\left(\mathrm{dd}, J_{\mathrm{HH}}=\right.$ $\left.7.6 \mathrm{~Hz}, J_{\mathrm{HH}}=1.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 11\right), 8.46\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H}_{7}\right), 13.88$ (br, $1 \mathrm{H}, \mathrm{OH}$ ). - ${ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=23.7(\mathrm{C} 9)$, 55.8 (C16), 62.1 (C8), 111.0 (C12), 117.5 (C3), 118.8 (C14), 119.4 (C5), 121.3 (C1), 127.4 (C10), 128.6 (C13), 131.8 (C15), 132.3 (C6), 132.6 (C4), 156.7 (C2), 161.9 (C11), 163.9 (C7). - MS (EI, 70 eV ): $m / z(\%)=255(35)[\mathrm{M}]^{+}$, 135 (100) $\left[\mathrm{CH}_{3} \mathrm{CHC}_{6} \mathrm{H}_{4} \mathrm{OMe}\right]^{+}, 105(5)\left[\mathrm{CH}_{3} \mathrm{CHC}_{6} \mathrm{H}_{5}\right]^{+}$. $\mathrm{C}_{16} \mathrm{H}_{17} \mathrm{NO}_{2}$ (255.32): calcd. C 75.27, H 6.71, N 5.49 ; found C 75.44, H 6.53, N 5.38 .

## (R)-N-(1-(3-Methoxyphenyl)ethyl)salicylaldimine (3)

Yield: $18.2 \mathrm{~g}(91 \%) .-[\alpha]^{20}\left(c=0.42, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right):-169^{\circ}$ $(578 \mathrm{~nm})$. $-\mathrm{IR}(\mathrm{KBr}): v=3053 \mathrm{~m}(\mathrm{H}-\mathrm{Ar}), 1624$ vs $(\mathrm{C}=\mathrm{N})$, 1576 vs (C=C) $\mathrm{cm}^{-1}$. - ${ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=$ $1.69\left(\mathrm{~d}, J_{\mathrm{HH}}=6.6 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H} 9\right), 3.87(\mathrm{~s}, 3 \mathrm{H}, \mathrm{H} 16), 4.58(\mathrm{q}$, $\left.J_{\mathrm{HH}}=6.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 8\right), 6.89\left(\mathrm{ddd}, J_{\mathrm{HH}}=7.2,6.8 \mathrm{~Hz}, J_{\mathrm{HH}}=\right.$ $1.6 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{H} 4,12), 6.95\left(\mathrm{~d}, J_{\mathrm{HH}}=6.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{H} 6,13\right)$,
$7.03\left(\mathrm{dd}, J_{\mathrm{HH}}=6.2 \mathrm{~Hz}, J_{\mathrm{HH}}=2.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 3\right), 7.31(\mathrm{dd}$, $\left.J_{\mathrm{HH}}=7.8 \mathrm{~Hz}, J_{\mathrm{HH}}=1.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 11\right), 7.39\left(\mathrm{ddd}, J_{\mathrm{HH}}=\right.$ $\left.6.8,6.5 \mathrm{~Hz}, J_{\mathrm{HH}}=1.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5\right), 8.45(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 7), 13.55$ (br, $1 \mathrm{H}, \mathrm{OH}$ ). - ${ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=25.3$ (C9), 55.6 (C16), 68.8 (C8), 112.7 (C13), 112.9 (C11), 117.4 (C3), 119.0 (C15), 119.2 (C5), 119.3 (C1), 130.1 (C14), 131.8 (C6), 132.7 (C4), 145.9 (C10), 160.3 (C2), 161.5 (C12), 163.9 (C7). - $\mathrm{C}_{16} \mathrm{H}_{17} \mathrm{NO}_{2}$ (255.32): calcd. C 75.27, H 6.71, N 5.49; found C 74.89, H 6.47, N 5.36.
(R)-N-(1-(4-Methoxyphenyl)ethyl)salicylaldimine (4)

Yield: $18.6 \mathrm{~g}(93 \%) .-[\alpha]^{20}\left(c=0.53, \mathrm{CHCl}_{3}\right):-170^{\circ}$ $(578 \mathrm{~nm}) .-\mathrm{IR}(\mathrm{KBr}): v=3054 \mathrm{~m}(\mathrm{H}-\mathrm{Ar}), 1626$ vs $(\mathrm{C}=\mathrm{N})$, 1609,1578 vs (C=C) $\mathrm{cm}^{-1} .-{ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=1.66\left(\mathrm{~d}, J_{\mathrm{HH}}=6.7 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H} 9\right), 3.85(\mathrm{~s}, 3 \mathrm{H}, \mathrm{H} 16), 4.57$ (q, $\left.J_{\mathrm{HH}}=6.7 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 8\right), 6.87-7.01(\mathrm{~m}, 4 \mathrm{H}, \mathrm{H} 3-6), 7.26-$ 7.39 (m, 4H, H11,12,14,15), 8.43 (s, 1H, H7), 13.58 (br, $1 \mathrm{H}, \mathrm{OH}) .-{ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}$ ): $\delta=1.56$ (d, $\left.J_{\mathrm{HH}}=6.8 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H} 9\right), 3.77(\mathrm{~s}, 3 \mathrm{H}, \mathrm{H} 16), 4.65\left(\mathrm{q}, J_{\mathrm{HH}}=\right.$ $6.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 8), 6.88-6.99$ (m, 4H, H3-6), $7.33-7.47$ (m, $4 \mathrm{H}, \mathrm{H} 11,12,14,15), 8.66$ (s, 1H, H7), 13.53 (br, 1H, OH). ${ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}$ ): $\delta=24.5$ (C9), 55.5 (C16), 66.6 (C8), 114.4 (C12,14), 116.8 (C3), 118.9 (C5), 119.1 (C1), 127.8 (C11,15), 132.0 (C6), 132.6 (C4), 136.3 (C10), 158.8 (C2), 160.9 (C13), 164.3 (C7). - MS (EI, 70 eV ): m/z $(\%)=255(85)[\mathrm{M}]^{+}, 135(100)\left[\mathrm{CH}_{3} \mathrm{CHC}_{6} \mathrm{H}_{4} \mathrm{OMe}\right]^{+}, 121$ (20) $\left[\mathrm{M}-\mathrm{CH}_{3} \mathrm{CC}_{6} \mathrm{H}_{4} \mathrm{OMe}\right]^{+}$, 105 (10) $\left[\mathrm{CH}_{3} \mathrm{CHC}_{6} \mathrm{H}_{5}\right]^{+}$. $\mathrm{C}_{16} \mathrm{H}_{17} \mathrm{NO}_{2}$ (255.32): calcd. C $75.27, \mathrm{H} 6.71, \mathrm{~N} 5.49$; found C 75.01, H 6.71, N 5.31.

## (R)-N-(1-(4-Bromophenyl)ethyl)salicylaldimine (5)

Yield: $22.0 \mathrm{~g}(92 \%) .-[\alpha]^{20}\left(c=0.61, \mathrm{CHCl}_{3}\right):-148^{\circ}$ $(578 \mathrm{~nm})$. $-\mathrm{IR}(\mathrm{KBr}): v=3049 \mathrm{~m}(\mathrm{H}-\mathrm{Ar}), 1616$ vs $(\mathrm{C}=\mathrm{N})$, 1575 vs (C=C) cm ${ }^{-1}$. - ${ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=$ $1.52\left(\mathrm{~d}, J_{\mathrm{HH}}=6.8 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H} 9\right), 4.43\left(\mathrm{q}, J_{\mathrm{HH}}=6.8 \mathrm{~Hz}, 1 \mathrm{H}\right.$, H8), $6.80\left(\right.$ ddd, $\left.J_{\mathrm{HH}}=7.4,6.4 \mathrm{~Hz}, J_{\mathrm{HH}}=1.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 4\right)$, $6.89\left(\mathrm{~d}, J_{\mathrm{HH}}=8.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 6\right), 7.16\left(\mathrm{dd}, J_{\mathrm{HH}}=6.2 \mathrm{~Hz}\right.$, $\left.J_{\mathrm{HH}}=1.8 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H} 3,11,15\right), 7.24\left(\mathrm{ddd}, J_{\mathrm{HH}}=6.8,7.0 \mathrm{~Hz}\right.$, $\left.J_{\mathrm{HH}}=1.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5\right), 7.40\left(\mathrm{dd}, J_{\mathrm{HH}}=4.8 \mathrm{~Hz}, J_{\mathrm{HH}}=\right.$ $1.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{H} 12,14), 8.32$ (s, 1H, H7), 13.22 (br, $1 \mathrm{H}, \mathrm{OH}$ ). ${ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}$ ): $\delta=1.56$ (d, $J_{\mathrm{HH}}=6.7 \mathrm{~Hz}$, $3 \mathrm{H}, \mathrm{H} 9), 4.69\left(\mathrm{q}, J_{\mathrm{HH}}=6.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 8\right), 6.93\left(\mathrm{ddd}, J_{\mathrm{HH}}=\right.$ $\left.7.8,6.6 \mathrm{~Hz}, J_{\mathrm{HH}}=1.0 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{H} 4,6\right), 7.38\left(\mathrm{ddd}, J_{\mathrm{HH}}=\right.$ $\left.7.7,6.4 \mathrm{~Hz}, J_{\mathrm{HH}}=1.7 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H} 3,11,15\right), 7.48\left(\mathrm{dd}, J_{\mathrm{HH}}=\right.$ $\left.6.4 \mathrm{~Hz}, J_{\mathrm{HH}}=1.7 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5\right), 7.57\left(\mathrm{dd}, J_{\mathrm{HH}}=6.7 \mathrm{~Hz}\right.$, $\left.J_{\mathrm{HH}}=1.7 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{H} 12,14\right), 8.69(\mathrm{~s}, 1 \mathrm{H}, \mathrm{H} 7), 13.28(\mathrm{br}, 1 \mathrm{H}$, $\mathrm{OH}) .-{ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}$ ): $\delta=24.5$ (C9), 66.6 (C8), 116.8 (C3), 119.1 (C13), 120.5 (C5), 128.9 (C11,15), 129.4 (C1), 131.8 (C12,14), 132.1 (C6), 132.8 (C4), 143.8 (C10), 160.7 (C2), 165.0 (C7). - MS (EI, 70 eV$): \mathrm{m} / \mathrm{z}(\%)=$ 304 (84) $[\mathrm{M}]^{+}$, $183(5)\left[\mathrm{CH}_{3} \mathrm{CHC}_{6} \mathrm{H}_{4} \mathrm{Br}\right]^{+}\left({ }^{79 / 81} \mathrm{Br}\right.$ isotopic pattern clearly visible for patterns following the 304 and


183 peaks, with masses given for the slightly more abundant ${ }^{79} \mathrm{Br}$-containing fragment), 121 (100) $\left[\mathrm{M}-\mathrm{CH}_{3} \mathrm{CC}_{6} \mathrm{H}_{4} \mathrm{Br}\right]^{+}$, 104 (55) $\left[\mathrm{CH}_{3} \mathrm{CHC}_{6} \mathrm{H}_{4}\right]^{+}, 77$ (10) $\left[\mathrm{C}_{6} \mathrm{H}_{5}\right]^{+} .-\mathrm{C}_{15} \mathrm{H}_{14} \mathrm{NOBr}$ (304.19): calcd. C 59.23, H $4.64, \mathrm{~N} 4.60$; found C 59.36 , H 4.61, N 4.55 .

## ( $R$ )-N-(1-(2-Naphthyl)ethyl)salicylaldimine (6)

Yield: $20.0 \mathrm{~g}(93 \%) .-[\alpha]^{20}\left(c=0.52, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right):-154^{\circ}$ $(578 \mathrm{~nm}),-173^{\circ}(546 \mathrm{~nm})$. $-\mathrm{IR}(\mathrm{KBr}): v=3048 \mathrm{~s}(\mathrm{H}-\mathrm{Ar})$, 1628 vs ( $\mathrm{C}=\mathrm{N}$ ), $1602 \mathrm{~s}, 1573$ vs ( $\mathrm{C}=\mathrm{C}$ ) $\mathrm{cm}^{-1}$. - ${ }^{1} \mathrm{H}$ NMR ( $\left.200 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}\right): \delta=1.69$ (d, $J_{\mathrm{HH}}=6.7 \mathrm{~Hz}, 3 \mathrm{H}$, $\mathrm{H} 9), 4.88\left(\mathrm{q}, J_{\mathrm{HH}}=6.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 8\right), 6.94\left(\mathrm{ddd}, J_{\mathrm{HH}}=7.7\right.$, $\left.8.2 \mathrm{~Hz}, J_{\mathrm{HH}}=1.0 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{H} 4,6\right), 7.36\left(\mathrm{ddd}, J_{\mathrm{HH}}=7.6\right.$, $\left.8.4 \mathrm{~Hz}, J_{\mathrm{HH}}=1.7 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5\right), 7.52$ (m, 3H, H3+nap), 7.60 $\left(\mathrm{dd}, J_{\mathrm{HH}}=8.5 \mathrm{~Hz}, J_{\mathrm{HH}}=1.7 \mathrm{~Hz}, 1 \mathrm{H}\right.$, nap $), 7.91-7.97(\mathrm{~m}$, 4 H , nap), 8.76 (s, 1H, H7), 13.55 (br, 1H, OH). - ${ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}$ ): $\delta=24.5$ (C9), 67.3 (C8), 116.8 (C3), 119.1 (C5), 119.2 (C1), 125.0 (C15), 125.3 (C16), 126.2 (C19), 126.6 (C12), 127.9 (C17), 128.2 (C14), 128.7 (C11), 132.2 (C6), 132.7 (C13), 132.8 (C4), 133.4 (C18), 141.9 (C10), 160.9 (C2), 164.9 (C7). - MS (EI, 70 eV ): m/z $(\%)=275(80)[\mathrm{M}]^{+}, 155(100)\left[\mathrm{CH}_{3} \mathrm{CHC}_{10} \mathrm{H}_{7}\right]^{+}, 121$ (20) $\left[\mathrm{M}-\mathrm{CH}_{3} \mathrm{CC}_{10} \mathrm{H}_{7}\right]^{+} .-\mathrm{C}_{19} \mathrm{H}_{17} \mathrm{NO}$ (275.35): calcd. C 82.88, H 6.22, N 5.09; found C 82.54, H 6.10, N 4.96.

## Cyclooctadiene- $\{(R)-N-(1-p h e n y l e t h y l)$ salicylaldiminato$\left.\kappa^{2} N, O\right\}-$-rhodium $(I)$ (7)

Two equivalents of ( $R$ )- $N$-(1-phenylethyl)salicylaldimine ( $80.4 \mathrm{mg}, 0.36 \mathrm{mmol}$ ) and one equivalent of $\left[\mathrm{Rh}\left(\mathrm{O}_{2} \mathrm{CMe}\right)\right.$ $\left(\eta^{4} \text {-cod) }\right]_{2}(96.3 \mathrm{mg}, 0.18 \mathrm{mmol})$ were dissolved in 10 mL of toluene $/ \mathrm{MeOH}(5: 1, \mathrm{v} / \mathrm{v})$ and the solution stirred for $5-6 \mathrm{~h}$ at r.t. The colour soon changed from red-orange to brightyellow. Then the solvent was evaporated in vacuo at $50^{\circ} \mathrm{C}$. The product was again dissolved in 10 mL of toluene $/ \mathrm{MeOH}$ ( $5: 1, \mathrm{v} / \mathrm{v}$ ), the solution stirred for 30 min and the solvent evaporated in vacuo. This procedure was repeated three times, and finally the yellow the product was dried in vacuo ( $0.1-0.2 \mathrm{mbar}$ ) at $60^{\circ} \mathrm{C}$. Single crystals suitable for X-ray measurements were grown by slow diffusion of diethyl ether into a chloroform solution of complex $\mathbf{7}$ after one week at r . t. Yield: $0.130 \mathrm{~g}(81 \%)$, based on $\left[\mathrm{Rh}\left(\mathrm{O}_{2} \mathrm{CMe}\right)\left(\eta^{4}-\operatorname{cod}\right)\right]_{2}$. UV/vis (7.109• $10^{-5} \mathrm{~mol} \mathrm{~mL}^{-1}, \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ): $\lambda_{\text {max }}\left(\lg \varepsilon_{\max }\right)=$ $392 \mathrm{~nm}(3.84), 234 \mathrm{~nm}(4.57) .-[\alpha]^{20}\left(c=0.26, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ : $+250^{\circ}(578 \mathrm{~nm}),+308^{\circ}(546 \mathrm{~nm}) .-[\alpha]^{20}\left(c=0.44, \mathrm{CHCl}_{3}\right)$ : $+182^{\circ}(578 \mathrm{~nm}) .-\operatorname{IR}(\mathrm{KBr}): v=3060,3030 \mathrm{w}(\mathrm{H}-\mathrm{Ar})$, 1626 sh (C=N), 1579 vs (C=C) cm ${ }^{-1}$. - ${ }^{1} \mathrm{H}$ NMR ( 200 MHz ,
[D $\mathrm{D}_{6}$ ]DMSO): $\delta=1.63$ (d, $J_{\mathrm{HH}}=6.9 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H} 9$ ), 1.87 (m, $4 \mathrm{H}, \mathrm{CH}_{2} \operatorname{cod}_{\text {exo }}$ ), $2.40\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{CH}_{2} \operatorname{cod}_{\text {endo }}\right), 3.77(\mathrm{~m}, 2 \mathrm{H}$, CHcod), $4.41\left(\mathrm{~m}, 3 \mathrm{H}, \mathrm{H}_{8}+\mathrm{CHcod}\right), 6.48\left(\mathrm{t}, J_{\mathrm{HH}}=7.4 / 6.3 \mathrm{~Hz}\right.$, $1 \mathrm{H}, \mathrm{H} 4), 6.64\left(\mathrm{~d}, J_{\mathrm{HH}}=8.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 6\right), 7.23\left(\mathrm{~d}, J_{\mathrm{HH}}=\right.$ $7.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{H} 3,5$ ), $7.28-7.39$ (m, 5H, H11-15), 8.13 (s, 1H, H7). - ${ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=1.58$ (d, $J_{\mathrm{HH}}=$ $6.9 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H} 9), 1.85\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{CH}_{2} \operatorname{cod}_{\mathrm{exo}}\right), 2.43(\mathrm{~m}, 4 \mathrm{H}$, $\mathrm{CH}_{2}$ cod $_{\text {endo }}$ ), 3.72 (m, 2H, CHcod), 4.37 (q, $J_{\mathrm{HH}}=6.8 \mathrm{~Hz}$, $1 \mathrm{H}, \mathrm{H} 8), 4.54(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CHcod}), 6.41$ (ddd, $J_{\mathrm{HH}}=6.8 \mathrm{~Hz}$, $\left.J_{\mathrm{HH}}=1.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 4\right), 6.77\left(\mathrm{~d}, J_{\mathrm{HH}}=8.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 6\right), 6.89$ $\left(\mathrm{dd}, J_{\mathrm{HH}}=6.0 \mathrm{~Hz}, J_{\mathrm{HH}}=1.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 3\right), 7.15-7.29(\mathrm{~m}$, $6 \mathrm{H}, \mathrm{H} 5,11,12,13,14,15), 7.82$ (d, $J_{\mathrm{HH}}=2.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 7$ ). ${ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=22.5(\mathrm{C} 9), 29.2,29.6$, $32.0,32.5\left(\mathrm{CH}_{2} \mathrm{cod}\right), 60.2(\mathrm{C} 8), 71.4\left(\mathrm{~d}, J_{\mathrm{CRh}}=14.6 \mathrm{~Hz}\right.$, CHcod), 73.5 (d, $\left.J_{\text {CRh }}=14.2 \mathrm{~Hz}, \mathrm{CHcod}\right), 85.3\left(\mathrm{~d}, J_{\mathrm{CRh}}=\right.$ $12.3 \mathrm{~Hz}, \mathrm{CHcod}), 85.7$ (d, $J_{\mathrm{CRh}}=12.1 \mathrm{~Hz}$, CHcod), 114.6 (C3), 119.7 (C5), 121.8 (C1), 127.7 (C13), 128.0 (C11,15), 129.0 (C12,14), 135.0 (C6), 135.5 (C4), 143.2 (C10), 165.4 (C2), 166.1 (C7). - MS (EI, 70 eV ): $m / z(\%)=435$ (86) $[\mathrm{M}]^{+}, 327$ (100) $[\mathrm{M}-\mathrm{cod}]^{+}, 225$ (16) $\left[\mathrm{HSB}^{*}\right]^{+}, 224$ (12) [SB] ${ }^{+}, 208$ (49) $\left[\mathrm{HSB}^{*}-\mathrm{OH}\right]^{+}, 206$ (35) $\left[\mathrm{SB}^{*}-\mathrm{H}_{2} \mathrm{O}\right]^{+}, 105$ (30) $\left[\mathrm{CH}_{3} \mathrm{CHC}_{6} \mathrm{H}_{5}\right]^{+}$, 103 (15) $[\mathrm{Rh}]^{+}, 77$ (7) $\left[\mathrm{C}_{6} \mathrm{H}_{5}\right]^{+}$. MS (CI, $\mathrm{NH}_{3}$ ): m/z $(\%)=436$ (100) $[\mathrm{M}+\mathrm{H}]^{+}, 327(10)$ $[\mathrm{M}-\mathrm{cod}]^{+}, 226$ (85) $\left[\mathrm{HSB}^{*}+\mathrm{H}\right]^{+}, 225$ (10) $\left[\mathrm{HSB}^{*}\right]^{+}$. $\mathrm{C}_{23} \mathrm{H}_{26} \mathrm{NORh}$ (435.37): calcd. C 63.45, H 6.02, N 3.22; found C 63.53, H 6.13, N 3.24 .

The same procedure was followed for the synthesis of the complexes $\mathbf{8 - 1 1}$ using the Schiff bases 2-6, respectively.

Cyclooctadiene- $\{(R)-N-(1-(2-m e t h o x y p h e n y l)$ ethyl $)$ salicyl-aldiminato- $\left.\kappa^{2} N, O\right\}$-rhodium $(I)$ (8)

Yield: $0.135 \mathrm{~g}(78 \%)$. - UV/vis $\left(8.526 \cdot 10^{-5} \mathrm{~mol}\right.$ $\left.\mathrm{mL}^{-1}, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right): \lambda_{\max }\left(\lg \varepsilon_{\max }\right)=388 \mathrm{~nm}(3.80), 236 \mathrm{~nm}$ (4.53). $-[\alpha]^{20}\left(c=0.25, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right):+200^{\circ}(578 \mathrm{~nm}),+220^{\circ}$ $(546 \mathrm{~nm}) .-\mathrm{IR}(\mathrm{KBr}): v=3044 \mathrm{w}(\mathrm{H}-\mathrm{Ar}), 1626 \mathrm{~s}(\mathrm{C}=\mathrm{N})$, 1573 vs ( $\mathrm{C}=\mathrm{C}$ ) $\mathrm{cm}^{-1}$. - ${ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=$ $1.54\left(\mathrm{~d}, J_{\mathrm{HH}}=6.9 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H} 9\right), 1.84\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{CH}_{2} \operatorname{cod}_{\mathrm{exo}}\right)$, 2.43 ( $\mathrm{m}, 4 \mathrm{H}, \mathrm{CH}_{2}$ cod $_{\text {endo }}$ ), 3.73 ( $\mathrm{m}, 1 \mathrm{H}, \mathrm{CHcod}$ ), 3.78 ( s , $\left.3 \mathrm{H}, \mathrm{H}_{16}\right), 4.29(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CHcod}), 4.42(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CHcod})$, $4.50(\mathrm{~m}, 1 \mathrm{H}, \mathrm{CHcod}), 4.63\left(\mathrm{q}, J_{\mathrm{HH}}=6.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 8\right)$, $6.38\left(\mathrm{ddd}, J_{\mathrm{HH}}=6.8 \mathrm{~Hz}, J_{\mathrm{HH}}=1.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 4\right), 6.76$ $\left(\mathrm{t}, J_{\mathrm{HH}}=9.5 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{H} 3,6\right), 6.85\left(\mathrm{dd}, J_{\mathrm{HH}}=6.2 \mathrm{~Hz}\right.$, $\left.J_{\mathrm{HH}}=1.7 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5\right), 6.93\left(\mathrm{~d}, J_{\mathrm{HH}}=7.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 11\right)$, $7.15\left(\mathrm{ddd}, J_{\mathrm{HH}}=6.8 \mathrm{~Hz}, J_{\mathrm{HH}}=1.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 13\right), 7.17-$ $7.27(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H} 12,14), 7.73\left(\mathrm{~d}, J_{\mathrm{HH}}=2.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 7\right)$. ${ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=22.8$ (C9), 28.9, 30.1, 31.7, $33.1\left(\mathrm{CH}_{2} \operatorname{cod}\right), 55.9\left(\mathrm{C}_{16}\right), 56.9\left(\mathrm{C}_{8}\right), 71.6\left(\mathrm{~d}, J_{\mathrm{CRh}}=\right.$ $14.5 \mathrm{~Hz}, \mathrm{CHcod}), 74.7$ (d, $J_{\mathrm{CRh}}=13.4 \mathrm{~Hz}$, CHcod), 84.0 (d, $\left.J_{C R h}=12.0 \mathrm{~Hz}, \mathrm{CHcod}\right), 85.0\left(\mathrm{~d}, J_{\mathrm{CRh}}=12.6 \mathrm{~Hz}, \mathrm{CHcod}\right)$, 111.4 (C12), 114.3 (C3), 119.8 (C14), 120.6 (C5), 121.7 (C1), 128.4 (C10), 129.6 (C13), 130.2 (C15), 134.6 (C6), 135.5 (C4), 157.1 (C2), 162.5 (C11), 165.9 (C7). - MS (EI,
$70 \mathrm{eV}): m / z(\%)=465(100)[\mathrm{M}]^{+}, 357(95)[\mathrm{M}-\mathrm{cod}]^{+}$, 327 (12) $[\mathrm{M}-\mathrm{cod}-\mathrm{HCHO}]^{+}, 255$ (5) $\left[\mathrm{HSB}^{*}\right]^{+}, 234$ (12) $\left[\mathrm{SB}-\mathrm{H}_{2} \mathrm{O}-\mathrm{H}_{2}\right]^{+}, 135(12)\left[\mathrm{CH}_{3} \mathrm{CHC}_{6} \mathrm{H}_{4} \mathrm{OMe}\right]^{+}, 103$ (5) $[R h]^{+} .-\mathrm{C}_{24} \mathrm{H}_{28} \mathrm{NO}_{2} \mathrm{Rh}(465.40)$ : calcd. C 61.94, H 6.06, N 3.01; found C 62.85, H 6.12, N 2.45 .

## Cyclooctadiene- $\{(R)-N-(1-(4-m e t h o x y p h e n y l) e t h y l)$ salicyl-aldiminato- $\left.\kappa^{2} N, O\right\}$-rhodium $(I)$ (9)

Yield: $0.130 \mathrm{~g}(75 \%)$. - UV/vis ( $1.398 \cdot 10^{-4} \mathrm{~mol}$ $\left.\mathrm{mL}^{-1}, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right): \lambda_{\max }\left(\lg \varepsilon_{\max }\right)=392 \mathrm{~nm}(3.70), 240 \mathrm{~nm}$ (4.38). $-[\alpha]^{20}\left(c=0.41, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right):+207^{\circ}(578 \mathrm{~nm}),+280^{\circ}$ $(546 \mathrm{~nm}) .-[\alpha]^{20}\left(c=0.56, \mathrm{CHCl}_{3}\right):+241^{\circ}(578 \mathrm{~nm}) .-\mathrm{IR}$ (KBr): $v=3062,3030 \mathrm{w}(\mathrm{H}-\mathrm{Ar}), 1624 \mathrm{~s}(\mathrm{C}=\mathrm{N}), 1577 \mathrm{vs}$ (C=C) $\mathrm{cm}^{-1}$. - ${ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}$ ): $\delta=$ $1.59\left(\mathrm{~d}, J_{\mathrm{HH}}=6.5 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H} 9\right), 1.88\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{CH}_{2} \operatorname{cod}_{\mathrm{exo}}\right)$, 2.42 ( $\mathrm{m}, 4 \mathrm{H}, \mathrm{CH}_{2} \mathrm{cod}_{\text {endo }}$ ), 3.74 ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{H} 16$ ), 3.76 (m, $2 \mathrm{H}, \mathrm{CHcod}), 4.34\left(\mathrm{q}, J_{\mathrm{HH}}=6.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 8\right), 4.43(\mathrm{~m}$, $2 \mathrm{H}, \mathrm{CHcod}), 6.47\left(\mathrm{t}, J_{\mathrm{HH}}=7.4 / 6.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 4\right), 6.63(\mathrm{~d}$, $\left.J_{\mathrm{HH}}=8.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 6\right), 6.94\left(\mathrm{~d}, J_{\mathrm{HH}}=8.8 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{H} 3,5\right)$, 7.25 (m, 4H, H11,12,14,15), 8.04 (s, 1H, H7). - ${ }^{1} \mathrm{H}$ NMR $\left(200 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=1.55\left(\mathrm{~d}, J_{\mathrm{HH}}=6.8 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H} 9\right)$, 1.88 ( $\mathrm{m}, 4 \mathrm{H}, \mathrm{CH}_{2} \mathrm{cod}_{\text {exo }}$ ), $2.42\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{CH}_{2} \mathrm{cod}_{\text {endo }}\right.$ ), 3.70 (m, 2H, CHcod), 3.73 ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{H} 16$ ), 4.53 (m, 2H, CHcod), $4.32\left(\mathrm{q}, J_{\mathrm{HH}}=6.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 8\right), 6.40\left(\mathrm{ddd}, J_{\mathrm{HH}}=6.8 \mathrm{~Hz}\right.$, $\left.J_{\mathrm{HH}}=1.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 4\right), 6.81(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H} 3,6), 6.89$ (ddd, $\left.J_{\mathrm{HH}}=6.1 \mathrm{~Hz}, J_{\mathrm{HH}}=1.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5\right), 7.15-7.22(\mathrm{~m}$, $4 \mathrm{H}, \mathrm{H} 11,12,14,15), 7.78\left(\mathrm{~d}, J_{\mathrm{HH}}=2.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 7\right)$. ${ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=22.7$ (C9), 29.1, 29.6, $32.0,32.5\left(\mathrm{CH}_{2} \mathrm{cod}\right), 55.7(\mathrm{C} 16), 59.7(\mathrm{C} 8), 71.2\left(\mathrm{~d}, J_{\mathrm{CRh}}=\right.$ $14.3 \mathrm{~Hz}, \mathrm{CHcod}), 73.6$ (d, $\left.J_{\text {CRh }}=14.0 \mathrm{~Hz}, \mathrm{CHcod}\right), 85.3$ (d, $\left.J_{\text {CRh }}=12.2 \mathrm{~Hz}, \mathrm{CHcod}\right), 85.7$ (d, $\left.J_{\text {CRh }}=12.2 \mathrm{~Hz}, \mathrm{CHcod}\right)$, 114.4 (C12,14), 114.6 (C3), 119.7 (C5), 121.8 (C1), 129.2 (C11,15), 134.9 (C6), 135.2 (C4), 135.5 (C10), 159.2 (C2), 165.2 (C13), 166.0 (C7). - MS (EI, 70 eV ): m/z (\%) = 465 (70) $[\mathrm{M}]^{+}, 357$ (100) $[\mathrm{M}-\mathrm{cod}]^{+}, 327$ (13) $[\mathrm{M}-\mathrm{cod}-$ $\mathrm{HCHO}]^{+}, 255$ (21) $\left[\mathrm{HSB}^{*}\right]^{+}, 238$ (41) $\left[\mathrm{HSB}^{*}-\mathrm{OH}\right]^{+}, 135$ (100) $\left[\mathrm{CH}_{3} \mathrm{CHC}_{6} \mathrm{H}_{4} \mathrm{OMe}\right]^{+}$, 105 (23) $\left[\mathrm{CH}_{3} \mathrm{CHC}_{6} \mathrm{H}_{5}\right]^{+}, 103$ (15) $[\mathrm{Rh}]^{+}, 77(10)\left[\mathrm{C}_{6} \mathrm{H}_{5}\right]^{+}$. - MS (CI, $\left.\mathrm{NH}_{3}\right): \mathrm{m} / \mathrm{z}(\%)=$ 466 (85) $[M+\mathrm{H}]^{+}, 256$ (100) $\left[\mathrm{HSB}^{*}+\mathrm{H}\right]^{+}, 135$ (20) $\left[\mathrm{CH}_{3} \mathrm{CHC}_{6} \mathrm{H}_{4} \mathrm{OMe}\right]^{+}$. - $\mathrm{C}_{24} \mathrm{H}_{28} \mathrm{NO}_{2} \mathrm{Rh}$ (465.40): calcd. C 61.94, H 6.06, N 3.01 ; found C 61.51, H $6.07, \mathrm{~N} 2.89$.

Cyclooctadiene- $\{(R)-N-(1-(4-b r o m o p h e n y l) e t h y l)$ salicylald-iminato- $\left.\kappa^{2} N, O\right\}$-rhodium (I) (10)

Yield: $0.150 \mathrm{~g}(79 \%)$. $\mathrm{UV} / \mathrm{vis}\left(7.408 \cdot 10^{-5} \mathrm{~mol} \mathrm{~mL}^{-1}\right.$, $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2}\right): \lambda_{\max }\left(\lg \varepsilon_{\max }\right)=394 \mathrm{~nm}(4.09), 244 \mathrm{~nm}(4.66)$. $[\alpha]^{20}\left(c=0.24, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right):+333^{\circ}(578 \mathrm{~nm}), 479^{\circ}(546 \mathrm{~nm})$. $[\alpha]^{20}\left(c=0.47, \mathrm{CHCl}_{3}\right):+308^{\circ}(578 \mathrm{~nm}) .-\mathrm{IR}(\mathrm{KBr}): v=$ 3045 w ( $\mathrm{H}-\mathrm{Ar}$ ), 1620 sh ( $\mathrm{C}=\mathrm{N}$ ), 1604 vs $(\mathrm{C}=\mathrm{C}) \mathrm{cm}^{-1}$. ${ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz},\left[\mathrm{D}_{6}\right] \mathrm{DMSO}$ ): $\delta=1.62$ (d, $J_{\mathrm{HH}}=$ $6.3 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H} 9$ ), 1.88 (m, 4H, CH2 $\mathrm{cod}_{\mathrm{exo}}$ ), $2.40(\mathrm{~m}, 4 \mathrm{H}$, $\mathrm{CH}_{2} \operatorname{cod}_{\text {endo }}$ ), $3.72(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CHcod}), 4.37\left(\mathrm{q}, J_{\mathrm{HH}}=6.8 \mathrm{~Hz}\right.$,

Table 2. Crystal structure data for $\mathbf{4}, \mathbf{5}$ and 7 .

|  | 4 | 5 | 7 |
| :---: | :---: | :---: | :---: |
| Formula | $\mathrm{C}_{16} \mathrm{H}_{17} \mathrm{NO}_{2}$ | $\mathrm{C}_{15} \mathrm{H}_{14} \mathrm{BrNO}$ | $\mathrm{C}_{23} \mathrm{H}_{26} \mathrm{NORh}$ |
| $M_{\text {r }}$ | 255.31 | 304.18 | 435.36 |
| Cryst. size $\left[\mathrm{mm}^{3}\right]$ | $0.42 \times 0.13 \times 0.12$ | $0.45 \times 0.21 \times 0.03$ | $0.39 \times 0.26 \times 0.12$ |
| Crystal system | orthorhombic | orthorhombic | monoclinic |
| Space group | $P 2{ }_{2} 1_{1}{ }_{1}$ | $P 2{ }_{1} 1_{1} 2_{1}$ | $P 2{ }_{1}$ |
| $a[\AA]$ | 5.724(2) | 5.8401(7) | 12.9992(16) |
| $b[\AA]$ | 12.633(5) | 7.6145(10) | 10.2131(13) |
| $c[\AA]$ | 19.237(7) | 31.146(4) | 14.6849(18) |
| $\beta$ [deg] | 90 | 90 | 102.961(2) |
| $V\left[\AA^{3}\right]$ | 1391.1(9) | 1385.0(3) | 1899.9(4) |
| Z | 4 | 4 | 4 |
| $D_{\text {calcd }}\left[\mathrm{g} \mathrm{cm}^{-3}\right]$ | 1.219 | 1.459 | 1.522 |
| $\mu\left(\mathrm{Mo} K_{\alpha}\right)\left[\mathrm{cm}^{-1}\right]$ | 0.80 | 29.55 | 9.10 |
| $F(000)$ [e] | 544 | 616 | 896 |
| $h k l$ range | $\pm 7 ; \pm 16 ; \pm 25$ | $\pm 7 ;-9,10 ;-42,41$ | $\pm 17 ; \pm 13 ; \pm 19$ |
| $((\sin \theta) / \lambda)_{\max }\left[\AA^{-1}\right]$ | 0.675 | 0.677 | 0.680 |
| Refl. measured | 12144 | 12448 | 17341 |
| Refl. unique | 1979 | 3358 | 8750 |
| $R_{\text {int }}$ | 0.0504 | 0.0449 | 0.0182 |
| Param. refined | 176 | 167 | 469 |
| $R(F) / w R\left(\mathrm{~F}^{2}\right)^{\text {a }}$ (all reflexions) | 0.0827/0.1116 | 0.0587/0.0607 | 0.0295/0.0465 |
| $x$ (Flack) | b | 0.017(9) | -0.006(16) |
| $\mathrm{GoF}\left(F^{2}\right)^{\mathrm{a}}$ | 1.037 | 0.844 | 0.936 |
| $\Delta \rho_{\text {fin }}(\mathrm{max} / \mathrm{min})\left[\mathrm{e} \AA^{-3}\right]$ | 0.164/-0.197 | 0.393/-0.352 | 0.388/-0.381 |

${ }^{\text {a }} R(F)=\left[\Sigma\left(\left\|F_{\mathrm{o}}|-| F_{\mathrm{c}}\right\|\right) / \Sigma\left|F_{\mathrm{o}}\right|\right] ; w R\left(F^{2}\right)=\left[\Sigma\left[w\left(F_{\mathrm{o}}{ }^{2}-F_{\mathrm{c}}^{2}\right)^{2}\right] / \Sigma\left[w\left(F_{0}{ }^{2}\right)^{2}\right]\right]^{1 / 2}$. - Goodness-of-fit $=\left[\Sigma\left[w\left(F_{\mathrm{o}}{ }^{2}-F_{\mathrm{c}}{ }^{2}\right)^{2}\right] /(n-p)\right]^{1 / 2}$. - Weighting scheme $w ; a / b=0.0601 / 0.0000$ for $\mathbf{4}, 0.0273 / 0.0000$ for 5 and $0.0201 / 0.0000$ for 7 with $w=1 /\left[\sigma^{2}\left(F_{0}^{2}\right)+(a P)^{2}+b P\right]$ where $P=\left(\max \left(F_{0}{ }^{2}\right.\right.$ or $\left.0)+2 F_{\mathrm{c}}^{2}\right) / 3 .-{ }^{\mathrm{b}}$ Anomalous scattering power is too small in combination with the data quality at hand to give a meaningful Flack parameter; Friedel opposites were therefore merged (MERG 4). The absolute configuration was established by the known absolute configuration of the starting amine.
$1 \mathrm{H}, \mathrm{H} 8), 4.42(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CHcod}), 6.49\left(\mathrm{t}, J_{\mathrm{HH}}=7.6 / 7.4 \mathrm{~Hz}\right.$, $1 \mathrm{H}, \mathrm{H} 4), 6.65\left(\mathrm{~d}, J_{\mathrm{HH}}=8.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 6\right), 7.28(\mathrm{~m}, 4 \mathrm{H}$, $\mathrm{H} 3,5,11,15), 7.58\left(\mathrm{~d}, J_{\mathrm{HH}}=8.2 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{H} 12,14\right), 8.12(\mathrm{~s}$, $1 \mathrm{H}, \mathrm{H} 7$ ). $-{ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=1.56$ (d, $J_{\mathrm{HH}}=$ $6.8 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H} 9), 1.88$ (m, 4H, CH2 $\operatorname{cod}_{\mathrm{exo}}$ ), $2.42(\mathrm{~m}, 4 \mathrm{H}$, $\mathrm{CH}_{2}$ cod $_{\text {endo }}$ ), $3.66(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CHcod}), 4.29\left(\mathrm{q}, J_{\mathrm{HH}}=6.8 \mathrm{~Hz}\right.$, $1 \mathrm{H}, \mathrm{H} 8), 4.55(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CHcod}), 6.42$ (ddd, $J_{\mathrm{HH}}=6.8 \mathrm{~Hz}$, $\left.J_{\mathrm{HH}}=1.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 4\right), 6.77\left(\mathrm{~d}, J_{\mathrm{HH}}=8.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 6\right)$, $6.90\left(\mathrm{dd}, J_{\mathrm{HH}}=6.2 \mathrm{~Hz}, J_{\mathrm{HH}}=1.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 3\right), 7.14(\mathrm{~d}$, $\left.J_{\mathrm{HH}}=8.5 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{H} 11,15\right), 7.20\left(\mathrm{ddd}, J_{\mathrm{HH}}=6.8 \mathrm{~Hz}, J_{\mathrm{HH}}=\right.$ $1.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5), 7.41\left(\mathrm{dd}, J_{\mathrm{HH}}=5.8 \mathrm{~Hz}, J_{\mathrm{HH}}=1.8 \mathrm{~Hz}\right.$, $2 \mathrm{H}, \mathrm{H} 12,14), 7.75$ (d, $J_{\mathrm{HH}}=2.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 7$ ). ${ }^{13} \mathrm{C}$ NMR $\left(50 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=21.1$ (C9), 27.9, 28.2, 30.7, 31.1 $\left(\mathrm{CH}_{2} \mathrm{cod}\right), 58.3(\mathrm{C} 8), 70.2\left(\mathrm{~d}, J_{\mathrm{CRh}}=14.1 \mathrm{~Hz}, \mathrm{CH} c o d\right)$, 72.0 (d, $\left.J_{\mathrm{CRh}}=14.6 \mathrm{~Hz}, \mathrm{CHcod}\right), 84.2\left(\mathrm{~d}, J_{\mathrm{CRh}}=12.2 \mathrm{~Hz}\right.$, CHcod), 84.5 (d, $\left.J_{\text {CRh }}=11.8 \mathrm{~Hz}, \mathrm{CHcod}\right), 113.4$ (C3), 118.2 (C13), 120.5 (C5), 128.1 (C1), 128.3 (C11,15), 130.8 (C12,14), 133.9 (C6), 134.2 (C4), 141.1 (C10), 164.1 (C2), 164.9 (C7). - MS (EI, 70 eV ): $m / z(\%)=513$ (81) [M] ${ }^{+}$, 405 (96) [M-cod] ${ }^{+}, 332(40)\left[\mathrm{M}-\mathrm{CH}_{3} \mathrm{CHC}_{6} \mathrm{H}_{4} \mathrm{Br}+\mathrm{H}_{2}\right]^{+}$, 303 (6) [ $\left.\mathrm{HSB}^{*}\right]^{+}, 223$ (14) [SB*-Br], 211 (15) [Rhcod] ${ }^{+}$, 184 (25) $\left[\mathrm{CH}_{3} \mathrm{CHC}_{6} \mathrm{H}_{4} \mathrm{Br}+\mathrm{H}\right]^{+}$, 104 (8) $\left[\mathrm{CH}_{3} \mathrm{CHC}_{6} \mathrm{H}_{4}\right]^{+}$, $103(5)[\mathrm{Rh}]^{+}$( ${ }^{79 / 81} \mathrm{Br}$ isotopic pattern clearly visible for patterns following the 513,405 , and 184 peaks, with masses
given for the slightly more abundant ${ }^{79} \mathrm{Br}$-containing fragment). - $\mathrm{C}_{23} \mathrm{H}_{25} \mathrm{BrNORh}$ (514.27): calcd. C 53.72, H 4.90, N 2.72; found C 53.21, H 5.00, N 2.51 .

Cyclooctadiene- $\{(R)-N-(1-(2-n a p h t h y l) e t h y l) s a l i c y l a l d i m i n-~$ ato- $\left.\kappa^{2} N, O\right\}$-rhodium (I) (11)

Yield: $0.145 \mathrm{~g}(81 \%)$. - UV/vis $\left(5.722 \cdot 10^{-5} \mathrm{~mol}\right.$ $\left.\mathrm{mL}^{-1}, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right): \lambda_{\max }\left(\lg \varepsilon_{\max }\right)=392 \mathrm{~nm}(4.17), 244 \mathrm{~nm}$ (4.78). $-[\alpha]^{20}\left(c=0.35, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right):+329^{\circ}(578 \mathrm{~nm}),+429^{\circ}$ $(546 \mathrm{~nm})$. - IR (KBr): $v=3053 \mathrm{w}, 3040 \mathrm{w}(\mathrm{H}-\mathrm{Ar}), 1622 \mathrm{vs}$ $(\mathrm{C}=\mathrm{N}), 1577$ vs ( $\mathrm{C}=\mathrm{C}$ ) $\mathrm{cm}^{-1}$. - ${ }^{1} \mathrm{H}$ NMR ( 200 MHz , $\mathrm{CDCl}_{3}$ ): $\delta=1.68\left(\mathrm{~d}, J_{\mathrm{HH}}=6.8 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H} 9\right), 1.89(\mathrm{~m}$, $4 \mathrm{H}, \mathrm{CH}_{2} \operatorname{cod}_{\text {exo }}$ ), $2.44\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{CH}_{2} \operatorname{cod}_{\text {endo }}\right.$ ), 3.77 ( $\mathrm{m}, 2 \mathrm{H}$, CHcod), $4.50\left(\mathrm{q}, J_{\mathrm{HH}}=6.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 8\right), 4.57(\mathrm{~m}, 2 \mathrm{H}$, CHcod), 6.35 (ddd, $J_{\mathrm{HH}}=6.5,6.1 \mathrm{~Hz}, J_{\mathrm{HH}}=1.0 \mathrm{~Hz}, 1 \mathrm{H}$, H4), $6.79\left(\mathrm{ddd}, J_{\mathrm{HH}}=7.8,7.1 \mathrm{~Hz}, J_{\mathrm{HH}}=1.5 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{H} 3,6\right)$, $7.16\left(\mathrm{ddd}, J_{\mathrm{HH}}=6.9,6.7 \mathrm{~Hz}, J_{\mathrm{HH}}=1.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 5\right), 7.38$ (ddd, $J_{\mathrm{HH}}=8.9,7.9 \mathrm{~Hz}, J_{\mathrm{HH}}=1.3 \mathrm{~Hz}, 2 \mathrm{H}$, nap), $7.42(\mathrm{~d}$, $\left.J_{\mathrm{HH}}=6.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{nap}\right), 7.72(\mathrm{~m}, 4 \mathrm{H}, \mathrm{nap}), 7.82\left(\mathrm{~d}, J_{\mathrm{HH}}=\right.$ $2.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H} 7$ ). ${ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=22.6$ (C9), 29.2, 29.6, 32.1, $32.6\left(\mathrm{CH}_{2} \mathrm{cod}\right), 60.4$ (C8), 71.4 (d, $\left.J_{\mathrm{CRh}}=14.2 \mathrm{~Hz}, \mathrm{CHcod}\right), 73.7$ (d, $\left.J_{\mathrm{CRh}}=14.3 \mathrm{~Hz}, \mathrm{CHcod}\right)$, 85.4 (d, $\left.J_{\mathrm{CRh}}=12.3 \mathrm{~Hz}, \mathrm{CHcod}\right), 85.8\left(\mathrm{~d}, J_{\mathrm{CRh}}=11.6 \mathrm{~Hz}\right.$,

CHcod), 114.7 (C3), 119.7 (C5), 121.8 (C1), 126.1 (C15), 126.6 (C16), 126.7 (C19), 126.8 (C12), 128.0 (C17), 128.5 (C14), 129.0 (C11), 133.0 (C6), 133.5 (C13), 135.1 (C4), 135.6 (C18), 140.7 (C10), 165.6 (C2), 166.1 (C7). - MS (EI, 70 eV$): m / z(\%)=485$ (64) $[\mathrm{M}]^{+}, 377$ (100) $[\mathrm{M}-$ cod $]^{+}, 275$ (5) $\left[\mathrm{HSB}^{*}\right]^{+}, 258$ (13) $\left[\mathrm{HSB}^{*}-\mathrm{OH}\right]^{+}, 155$ (7) $\left[\mathrm{CH}_{3} \mathrm{CHC}_{10} \mathrm{H}_{7}\right]^{+} .-\mathrm{C}_{27} \mathrm{H}_{28} \mathrm{NORh}$ (485.43): calcd. C 66.81, H 5.81, N 2.89; found C 66.71, H 6.45, N 2.38 .

## X-Ray structure determinations

Data Collection: Bruker AXS with CCD area detector, temperature $203(2) \mathrm{K}, \operatorname{Mo} K_{\alpha}$ radiation $(\lambda=0.71073 \AA)$, graphite monochromator, $\omega$ scans, data collection and cell refinement with SMART [47], data reduction with SAINT [47], experimental absorption correction with SADABS [48].

Structure Analysis and Refinement: The structure was solved by Direct Methods (SHELXS-97) [49], refinement was carried out by full-matrix least-squares on $F^{2}$ using the SHELXL-97 program suite [49]. All non-hydrogen positions were found and refined with anisotropic temperature factors. Hydrogen atoms on oxygen $(-\mathrm{OH})$ were found and fully refined in $\mathbf{4}$ and 5. Hydrogen atoms on C (phenyl, $\mathrm{CH}, \mathrm{CH}_{2}$ and
$\mathrm{CH}_{3}$ ) were calculated with appropriate riding models (AFIX $43,13,23$ and 33 , respectively) and $U_{\text {eq }}(\mathrm{H})=1.2 U_{\text {eq }}(\mathrm{CH})$ or $U_{\text {eq }}(\mathrm{H})=1.5 U_{\text {eq }}\left(\mathrm{CH}_{3}\right)$. Details of the X-ray structure determinations and refinements are provided in Table 2. Graphics were drawn with DIAMOND (Version 3.1c) [50]. Computations on the supramolecular interactions were carried out with PLATON for Windows [15].

CCDC 636583 for 4, 636584 for 5, and 636585 for 7 contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

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[1] A.E.E. Amr, M. Abo-Ghalia, M. M. Abdalah, Z. Naturforsch. 2006, 61b, 1335; R. O. Doroschuk, E. G. Petkova, R. D. Lampeka, K. V. Domasevitch, M. V. Gorichko, Z. Naturforsch. 2006, 61b, 1361; W. Hoffmüller, H. Dialer, W. Beck, Z. Naturforsch. 2005, 60b, 1278; R. Urban, W. Beck, Z. Naturforsch. 2005, 60b, 1071; S. Klingelhöfer, M. Wiebcke, P. Behrens, Z. Anorg. Allg. Chem. 2007, 633, 113; T. Ederer, R. S. Herrick, W. Beck, Z. Anorg. Allg. Chem. 2007, 633, 235; B. Wisser, C. Janiak, Z. Anorg. Allg. Chem. 2007, in press; T. Hauck, K. Sünkel, W. Beck. Z. Anorg. Allg. Chem. 2006, 632, 2305; J.-P. Li, J.-S. Zhao, Z. Anorg. Allg. Chem. 2006, 632, 1897; R. Urban, W. Beck, Z. Anorg. Allg. Chem. 2006, 632, 955; L.F. Ma, L. Y. Wang, J. G. Wang, Y. F. Wang, X. Feng, Z. Anorg. Allg. Chem. 2006, 632, 487; O. Seewald, U. Florke, H. Egold, H. J. Haupt, M. Schwefer, Z. Anorg. Allg. Chem. 2006, 632, 204; H. Brunner, C. Keck, Z. Anorg. Allg. Chem. 2005, 631, 2555; T. J. Colacot, N. S. Hosmane, Z. Anorg. Allg. Chem. 2005, 631, 2659; R. Urban, D. Veghini, H. Berke, W. Beck, Z. Anorg. Allg. Chem. 2005, 631, 2715; G. Müller, J. Brand, Z. Anorg. Allg. Chem. 2005, 631, 2820; K. Lappalainen, K. Yliheikkila, A. S. Abu-Surrah, M. Polamo, M. Leskela, T. Repo, Z. Anorg. Allg. Chem. 2005, 631, 763; B. Paul, C. Näther, K. M. Fromm, C. Janiak, CrystEngComm. 2005, 7, 309; G. Vujevic, C. Janiak, Z. Anorg. Allg. Chem. 2003, 629, 2585.
[2] X. G. Zhou, J. Zhao, A. M. Santos, F. E. Kühn, Z. Naturforsch. 2004, 59b, 1223; D. Koch, W. Hoffmüller, K. Polborn, W. Beck, Z. Naturforsch. 2001, 56b, 403.
[3] C.-M. Chee, J.-S. Huang, Coord. Chem. Rev. 2003, 242, 97; P. G. Cozzi, Chem. Soc. Rev. 2004, 33, 410.
[4] E. D. McKenzie, S. J. Selvey, Inorg. Chim. Acta 1985, 101, 127; T. Akitsu, Y. Einaga, Acta Cryst. 2004, C60, m640; L. Z. Flores-Lopez, M. Parra-Hake, R. Somanathan, P.J. Walsh, Organometallics 2000, 19, 2153.
[5] H. Brunner, M. Niemetz, M. Zabel, Z. Naturforsch. 2000, 55b, 145; H. Sakiyama, H. Okawa, N. Matsumoto, S. Kida, J. Chem. Soc., Dalton Trans. 1990, 2935; H. Sakiyama, H. Okawa, N. Matsumoto, S. Kida, Bull. Chem. Soc. Jpn. 1991, 64, 2644; C. Evans, D. Luneau, J. Chem. Soc., Dalton Trans. 2002, 83; J. M. Femandez, O. L. Ruiz-Ramirez, R. A. Toscano, N. Macias-Ruvalcaba, M. Aguilar-Martinez, Transition Met. Chem. 2000, 25, 511.
[6] Asymmetric hydrogenation of acetophenone: I. Karamé, M. L. Tommasino, R. Faure, M. Lemaire, Eur. J. Org. Chem. 2003, 1271; I. Karamé, M. Jahjah, A. Messaoudi, M. L. Tommasino, M. Lemaire, Tetrahedron: Asymmetry 2004, 15, 1569.
[7] Asymmetric epoxidation of olefins: P. Guo, K.-Y. Wong, Electrochem. Commun. 1999, 1, 559.
[8] Asymmetric trimethylsilylcyanation of aromatic aldehydes: Z.-H. Yang, L.-X. Wang, Z.-H. Zhou, Q.-L.

Zhou, C.-C. Tang, Tetrahedron: Asymmetry 2001, 12, 1579.
[9] Dioxomolybdenum and oxovanadium complexes with bi-, tri- and tetra-dentate chiral Schiff bases: C. Zhang, G. Rheinwald, V. Lozan, B. Wu, P.-G. Lassahn, H. Lang, C. Janiak, Z. Anorg. Allg. Chem. 2002, 628, 1259; S. P. Rath, T. Ghosh, S. Mondal, Polyhedron 1997, 16, 4179; G. Santoni, D. Rehder, J. Inorg. Biochem. 2004, 98, 758.
[10] H. Brunner, R. Oeschey, B. Nuber, J. Chem. Soc., Dalton Trans. 1996, 1499.
[11] H. Brunner, T. Zwack, M. Zabel, W. Beck, A. Boehm, Organometallics 2003, 22, 1741.
[12] S. K. Mandal, A. R. Chakravarty, J. Organomet. Chem. 1991, 417, C59.
[13] S. K. Mandal, A. R. Chakravarty, J. Chem. Soc., Dalton Trans. 1992, 1627; S. K. Mandal, A.R. Chakravarty, Inorg. Chem. 1993, 32, 3851.
[14] H. Brunner, H. Fisch, J. Organomet. Chem. 1987, 335, 1; H. Brunner, H. Leyerer, J. Organomet. Chem. 1987, 334, 369.
[15] R. Bonnaire, C. Potvin, J. M. Manoli, Inorg. Chim. Acta 1980, 45, L255.
[16] H. Brunner, A. F. M. M. Rahman, Z. Naturforsch. 1983, 38b, 1332.
[17] C. A. Rogers, B. O. West, J. Organomet. Chem. 1974, 70, 445.
[18] R. Bonnaire, J. M. Manoli, C. Potvin, N. Platzer, N. Goasdoue, D. Davoust, Inorg. Chem. 1982, 21, 2032.
[19] R. J. Cozens, K. S. Murray, B. O. West, J. Organomet. Chem. 1971, 27, 399.
[20] N. Platzer, N. Goasdoue, R. Bonnaire, J. Organomet. Chem. 1978, 160, 455.
[21] M. Enamullah, M. Hasegawa, T. Hoshi, Abstract, Bangladesh Chem. Soc. Conf., Jahangirnagar University (Bangladesh) 2003; M. Enamullah, M. Hasegawa, J. Okubo, T. Hoshi, Abstract, Bangladesh Chem. Soc. Conf., Dhaka University (Bangladesh) 2004.
[22] M. Enamullah, M. Hasegawa, J. Okubo, T. Hoshi, J. Bangladesh Chem. Soc. 2005, 18, 165; M. Enamullah, M. Uddin, W. Linert, J. Coord. Chem., in press.
[23] M. Enamullah, A. Sharmin, M. Hasegawa, T. Hoshi, A. C. Chamayou, C. Janiak, Eur. J. Inorg. Chem. 2006, 2146.
[24] M. Enamullah, M. Uddin, K. S. Hagen, C. Janiak, to be submitted.
[25] J. G. Leipoldt, E. C. Grobler, Inorg. Chim. Acta 1983, 72, 17.
[26] A.P. Martinez, M. P. Garcia, F. J. Lahoz, L. A. Oro, Inorg. Chem. Commun. 2002, 5, 245.
[27] G. S. Rodman, K. R. Mann, Inorg. Chem. 1988, 27, 3338.
[28] C. Tejel, M. A. Ciriano, M. Bordonaba, J. A. Lopez, F. J. Lahoz, L. A. Oro, Inorg. Chem. 2002, 41, 2348.
[29] J. C. Bayon, G. Net, P. G. Rasmussen, J. B. Kolowich, J. Chem. Soc., Dalton Trans. 1987, 3003.
[30] R. Bonnaire, J. M. Manoli, N. Potvin, N. Platzer, N. Goasdoue, Inorg. Chem. 1981, 20, 2691.
[31] R. Ugo, G. La Monica, S. Cenini, F. Bonati, J. Organomet. Chem. 1968, 11, 159.
[32] J. Kriz, K. Bouchal, J. Organomet. Chem. 1974, 64, 255.
[33] R. Uson, L.A. Oro, M.A. Ciriano, R.J. Gonzales, J. Organomet. Chem. 1981, 205, 259.
[34] S.L. James, D. M. P. Mingos, X. Xu, A.J.P. White, D. J. Williams, J. Chem. Soc., Dalton Trans. 1998, 1335.
[35] R. Aumann, I. Goettker-Schnetmann, R. Froehlich, P. Saarenketo, C. Holst, Chem. Eur. J. 2001, 7, 711.
[36] H. H. Monfared, O. Pouralimardan, C. Janiak, Z. Naturforsch. 2007, 62, in press.
[37] C. Janiak, J. Chem. Soc., Dalton Trans. 2000, 3885.
[38] T. Dorn, C. Janiak, K. Abu-Shandi, CrystEngComm. 2005, 7, 633; K. Abu-Shandi, H. Winkler, H. Paulsen, R. Glaum, B. Wu, C. Janiak, Z. Anorg. Allg. Chem. 2005, 631, 2705; S. Banerjee, A. Ghosh, B. Wu, P.-G. Lassahn, C. Janiak, Polyhedron 2005, 24, 593; S. Banerjee, B. Wu, P.-G. Lassahn, C. Janiak, A. Ghosh, Inorg. Chim. Acta 2005, 358, 535; C. Zhang, G. Rheinwald, V. Lozan, B. Wu, P.-G. Lassahn, H. Lang, C. Janiak, Z. Anorg. Allg. Chem. 2002, 628,1259 ; E. Craven, E. Mutlu, D. Lundberg, S. Temizdemir, S. Dechert, H. Brombacher, C. Janiak, Polyhedron 2002, 21, 553.
[39] X.-J. Yang, F. Drepper, B. Wu, W.-H. Sun, W. Haehnel, C. Janiak, Dalton Trans. 2005, 256, and supplementary material therein.
[40] M. Nishio, CrystEngComm. 2004, 6, 130; M. Nishio, M. Hirota, Y. Umezawa, The $C H / \pi$ interaction, WileyVCH, New York, 1998; Y. Umezawa, S. Tsuboyama, K. Honda, J. Uzawa, M. Nishio, Bull. Chem. Soc. Jpn. 1998, 71, 1207.
[41] C. Janiak, S. Temizdemir, S. Dechert, W. Deck, F. Girgsdies, J. Heinze, M. J. Kolm, T. G. Scharmann, O. M. Zipffel, Eur. J. Inorg. Chem. 2000, 1229.
[42] T. Dorn, A.-C. Chamayou, C. Janiak, New. J. Chem. 2006, 30, 156.
[43] G. R. Desiraju, T. Steiner, The Weak Hydrogen Bond in Structural Chemistry and Biology, IUCr Monograph on Crystallography, Vol. 9, Oxford Science, Oxford, 1999.
[44] Based on a search of the Cambridge Structure Database (CSD), Version 5.27 (November 2005) with 2 updates (January 2006, May 2006).
[45] G. Giordano, R. H. Crabtree, Inorg. Synth. 1990, 28, 88.
[46] Z. Nagy-Magos, S. Vastag, B. Heil, L. Marko, J. Organomet. Chem. 1979, 171, 97; Z. Nagy-Magos, P. Kvintovics, L. Marko, Trans. Met. Chem. 1980, 5, 186.
[47] SMART, Data Collection Program for the CCD AreaDetector System; SAINT, Data Reduction and Frame Integration Program for the CCD Area-Detector System. Bruker Analytical X-ray Systems Inc., Madison, Wisconsin (USA) 1997.
[48] G. Sheldrick, SADABS, Area-detector absorption correction, University of Göttingen, Göttingen (Germany) 1996.
[49] G. M. Sheldrick, SHELXS/L-97, Programs for Crystal

Structure Analysis, University of Göttingen, Göttingen (Germany) 1997.
[50] K. Brandenburg, DIAMOND (Version 3.1c), Crystal and Molecular Structure Visualization, Crystal Impact, K. Brandenburg \& H. Putz Gbr, Bonn (Germany) 2004.
[51] A. L. Spek, Platon, A. Multipurpose Crystallographic Tool, Utrecht University, Utrecht (The Netherlands) 2000. See also: A.L. Spek, Acta Crystallogr. 1990, A46, C34. Windows implementation: L. J. Farrugia, Version 80205, University of Glasgow, Glasgow, Scotland (U. K.) 2005.
[52] H. D. Flack, Acta Crystallogr. 1983, A39, 876.

