

**Polysulfonylamine, CLXXX [1].**

**Bis(triphenylphosphoranylidene)ammonium-di(methansulfonyl)amid:  
Eine Kristallstruktur mit CH/ $\pi$ -Wechselwirkungen zwischen Methyl-  
und Phenylgruppen**

Polysulfonylamines, CLXXX [1]. Bis(triphenylphosphoranylidene)ammonium  
Di(methanesulfonyl)amide: A Crystal Structure Featuring Methyl-Phenyl CH/ $\pi$  Interactions

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The structure of  $[\text{Ph}_3\text{PNPPh}_3][(\text{MeSO}_2)_2\text{N}]$  (monoclinic,  $P2_1/c$ ,  $Z' = 1$ ; single-crystal X-ray diffraction at  $-140^\circ\text{C}$ ) displays chains of relatively small anions, surrounded by four parallel columns of bulky cations, giving a rectangular array. In contrast to a previously reported series of onium dimesylamides with smaller cations, interanion hydrogen bonding  $\text{C}_{\text{Me}}\text{--H}\cdots\text{O}$ , originating from the inductively activated methyl groups, is not observed in the present structure. In consequence, the packing arrangement is determined by multiple phenyl embraces between cations, weak cation-anion hydrogen bonds  $\text{C}_{\text{Ph}}\text{--H}\cdots\text{O/N}$  and, most notably, by methyl-phenyl CH/ $\pi$  interactions between anions and cations. One methyl group forms two  $\text{C--H}\cdots\pi$  contacts to different phenyl rings (hydrogen-to-centroid distances: 272 and 285 pm), the other lies in a tripodal mode above an aromatic ring plane (hydrogen-to-centroid distances: *ca.* 330 pm).

*Key words:* Activated Methyl Groups, Methyl-Phenyl CH/ $\pi$  Interactions, Weak Hydrogen Bonding,  
Multiple Phenyl Embraces, Sulfonamides