## Kristallstruktur und Eigenschaften von Chrom(II)-Tetrachloroaluminat

Crystal Structure and Properties of Chromium(II)-Tetrachloroaluminate

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Chromium(II)tetrachloroaluminate can be prepared by reaction of chromium, chromium trichloride and aluminum trichloride in the molar ratio 1:1.94:5.79 or by reaction of stoichiometric amounts of chromium dichloride and aluminum trichloride. Sublimation of the compound yields phase-pure samples of colourless crystals. Cr(AlCl<sub>4</sub>)<sub>2</sub> crystallizes in the non-centrosymmetric orthorhombic space group  $Pca2_1$  (No. 29), a = 1511.38(3), b = 604.71(1), c = 1301.76(3) pm. The aggregation of Cr(AlCl<sub>4</sub>)<sub>2</sub> molecules leads to a square-bipyramidal coordination of the chromium atom typical of high-spin  $d^4$  ions due to the Jahn-Teller effect. The relationship to centrosymmetric Pd(AlCl<sub>4</sub>)<sub>2</sub> ( $P2_1/c$ ) which is built up of similar molecules was established by means of a group-subgroup symmetry tree. The temperature dependence of the magnetic susceptibility agrees with Curie-Weiss behaviour ( $\mu_{eff} = 4.70 \ \mu_{B}$ ,  $\Theta = -1.3 \ K$ ).

Key words: Chromium, Tetrachloroaluminate, Crystal Structure, Jahn-Teller Distortion, Group-Subgroup Relations