Gemischte Erdalkalimetall-Trielide  $A^{II}M1_x^{III}M2_{2-x}^{III}$  ( $A^{II}$  = Ca, Sr, Ba;  $M^{III}$  = Al, Ga, In). Strukturchemische und bindungstheoretische Untersuchungen

Mixed Alkaline Earth Trielides  $A^{II}M1_x^{III}M2_{2-x}^{III}$  ( $A^{II}$  = Ca, Sr, Ba;  $M^{III}$ =Al, Ga, In). A Structural and Theoretical Study

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The binary alkaline earth trielides of the composition  $A^{\rm II}M_2^{\rm III}$  exhibit a puzzling variety of structure types ranging from electron precise Zintl compounds like CaIn<sub>2</sub> and KHg<sub>2</sub> (both with networks of four-bonded  $M^-$  entities) and the AlB<sub>2</sub> structure type (with graphite analogue M sheets) to the cubic Laves phases e.g. of CaAl<sub>2</sub>. The examination of the phase stabilities of mixed compounds  $AM1_{1}^{11}M2_{2}^{11}$  of two trielides allows to separate the stability ranges in a structure map by taking the electronegativity differences of  $M^{\rm III}$  and  $A^{\rm II}$  ( $\Delta \rm EN$ ) and the radius ratios (RR =  $r_{\rm M}/r_{\Delta}$ ) into account: The CaIn<sub>2</sub>-type is stable at comparatively large RR, for example over the whole range CaGa<sub>2</sub> – CaIn<sub>2</sub> and even up to CaAl<sub>0.6</sub>Ga<sub>1.4</sub> and CaAl<sub>1.2</sub>In<sub>0.8</sub>, and in SrIn<sub>2</sub>, together with a limited substitution of In by Al or Ga. The KHg<sub>2</sub>-type is observed in a region of lower RR: In BaIn<sub>2</sub>, a substitution of In by 50 % Al and 30 % Ga is possible without a general structure change, in SrAl<sub>2</sub> this holds for a content of up to 50 % In. At high ΔEN and low RR values (e, g, Sr/Ba-Ga), the ideal AlB<sub>2</sub> structure type exhibits a distinct stability range; only for small RR around CaAl2 the MgCu2-type is stable. FP-LAPW band structure calculations of the binary trielides allow to explain the structural changes qualitatively. In the case of the electron precise phases forming the CaIn<sub>2</sub>, KHg<sub>2</sub> or AlB<sub>2</sub> structure type, details of the bonding situation (such as M-M distances) as well as differences to other isoelectronic compounds can be rationalized taking the incomplete charge transfer from the alkaline earth towards the triel elements into account. This causes a partial depopulation of some of the bonding and a population of predominantly antibonding states.

Key words: Trielides, Alkaline Earth, Band Structure Calculation, Structure Map