Synthesis, Crystal Structure and Thermal Decomposition Mechanism of a Dysprosium(III) *p*-Fluorobenzoate 1,10-Phenanthroline Complex

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A dinuclear dysprosium(III) p-fluorobenzoate 1,10-phenanthroline complex, $[Dy(p\text{-FBA})_3\text{phen}]_2$ was synthesized and characterized by elemental analysis, UV and IR spectroscopy, single crystal X-ray diffraction, molar conductance, and TG-DTG techniques. It crystallizes in the triclinic space group $P\bar{1}$ with a=9.895(5), b=11.754(6), c=14.756(10) Å; $\alpha=106.660(9)^\circ$, $\beta=107.956(9)^\circ$, $\gamma=101.472(7)^\circ$; Z=1. The Dy(III) ions are eight coordinate including one terminal bidentate chelating carboxylate group, four bridging carboxylate groups and one 1,10-phenanthroline molecule. The thermal decomposition of $[Dy(p\text{-FBA})_3\text{phen}]_2$ has been followed by thermal analysis. The lifetime equation at weight-loss of 10 % was deduced as $\ln \tau = -27.0798 + 19010.2434/T$ by isothermal thermogravimetric analysis.

Key words: p-Fluorobenzoic Acid, Crystal Structure, 1,10-Phenanthroline, Dysprosium Complex, Thermal Analysis