

# Hydrothermal Syntheses and Crystal Structures of Two New Polyoxometalate-based Charge Transfer Salts

Chunhua Zhang<sup>a,b</sup>, Fengyan Li<sup>a</sup>, and Lin Xu<sup>a</sup>

<sup>a</sup> Key Laboratory of Polyoxometalates Science of Ministry of Education, College of Chemistry, Northeast Normal University, Changchun 130024, P. R. China

<sup>b</sup> Department of Chemistry, Baicheng Teacher's College, Baicheng 137000/P.R. China

Reprint requests to Prof. Dr. L. Xu. Fax: +86-431-/5099668. E-mail: linxu@nenu.edu.cn

Z. Naturforsch. **61b**, 1377 – 1382 (2006); received April 4, 2006

Two new charge transfer salts based on Keggin polyoxoanions,  $[\text{Hen}]_2[\text{H}_2\text{en}][\text{PMo}_{11}^{\text{VI}}\text{Mo}^{\text{V}}\text{O}_{40}] \cdot 3.5 \text{H}_2\text{O}$  (**1**) and  $[\text{Hpy}]_4[\text{GeMo}_{12}^{\text{VI}}\text{O}_{40}] \cdot 2\text{H}_2\text{O}$  (**2**) (en = ethylenediamine, py = pyridine), have been synthesized and characterized by single crystal X-ray diffraction, elemental analyses, IR spectra, UV/vis spectra, cyclic voltammograms and X-ray photoelectron spectra. The IR spectra and solid reflectance electronic spectra of the title compounds indicate that interactions exist between the polyanions and the organic substrates, which are probably caused by charge transfer.

*Key words:* Polyoxometalates, Charge Transfer Salts, Hydrothermal Synthesis, Crystal Structure