

# Cyclisierung von Di(*tert*-butyl-methyl)ketazin zu 1,2-Diaza-3-bora- und 1,2-Diaza-3-sila-cyclopent-5-enen

Cyclization of Di(*tert*-butyl-methyl)ketazine to 1,2-Diaza-3-bora- and 1,2-Diaza-3-sila-cyclopent-5-enes

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Z. Naturforsch. **61b**, 1261 – 1274 (2006); eingegangen am 24. April 2006

The results of quantum chemical calculations on lithium ketazides suggest mainly four isomeric structures with different modes of lithium coordination (**A – D**). A monolithium ketazide thf-adduct (**1**) was isolated supporting the results of the quantum chemical calculations. In reactions of the lithiated di(*tert*-butyl-methyl)ketazine with  $\text{BCl}_3$  and  $\text{Cl}_2\text{BPh}$ , 1,2-aza-azonia-3-borata-cyclopent-5-enes (**2, 3**) were isolated. Substitution of a chlorine atom of **2** and **3** with *t*-BuLi leads to the formation of derivatives **4** and **5**. HCl elimination from **2** with  $\text{Et}_3\text{N}$  gives – *via* a diazaboracyclopentene (**6**) – a bicyclus **7**. In the reaction of the dilithiated ketazine with  $\text{F}_2\text{BN}(\text{SiMe}_3)_2$ , the diaza-boracyclopentene **8** is obtained while with  $\text{Cl}_4\text{Si}$ ,  $\text{F}_3\text{SiN}(\text{SiMe}_3)_2$ , and  $\text{Cl}_2\text{SiMe}_2$  the diazasilacyclopentenes **9 – 11** are generated.  $\text{SiF}_4$  reacts with the dilithium ketazide to give a spirocyclus (**12**). The monolithium ketazide and  $\text{Cl}_2\text{SiMe}_2$  react at 30 °C to give a four-membered ring isomer of the substitution product which is formed *via* a 1,3-chlorine shift from silicon to carbon (**13**). A tetrameric silanolate was isolated as a by-product in this reaction. It gives evidence for the structure of lithium ketazide **A**. Crystal structures of **5, 7, 10**, and **14** are reported.

**Key words:** Lithium Ketazides, Quantum Chemical Calculations, Diaza-bora-cyclopentenes, Diaza-sila-cyclopentenes, Aza-silacyclobutane