

Polysulfonylamine, CLXXV [1]. Intermolekulare Wechselwirkungen in kristallinen Di(organosulfonyl)aminen. Teil 3 [2]. Supramolekulare Strukturen von Di(benzolsulfonyl)amin, (Benzolsulfonyl)(methansulfonyl)amin und Di(ethansulfonyl)amin

Polysulfonylamines, CLXXV [1]. Intermolecular Interactions in Crystalline Di(organosulfonyl)-amines. Part 3 [2]. Supramolecular Structures of Di(benzenesulfonyl)amine, (Benzenesulfonyl)-(methanesulfonyl)amine, and Di(ethanesulfonyl)amine

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Low-temperature X-ray structures of the following di(organosulfonyl)amines, $\text{HN}(\text{SO}_2\text{R})(\text{SO}_2\text{R}')$, are compared in order to study the packing effects of group-specific intermolecular interactions: $\text{R} = \text{R}' = \text{phenyl}$ (**1**, monoclinic, $P2_1/c$, $Z' = 1$); $\text{R} = \text{phenyl}$, $\text{R}' = \text{methyl}$ (**2**, monoclinic, $P2_1/n$, $Z' = 1$); $\text{R} = \text{R}' = \text{ethyl}$ (**3**, monoclinic, $P2_1/c$, $Z' = 1$). The molecules of each compound are associated into catemers by $\text{N-H}\cdots\text{O}$ hydrogen bonds of graph set C(4). These catemers form molecular layers *via* distinctive cross-linking modes, which underline the specific propensity of the organic groups to establish directional intermolecular interactions. Thus, the packing of **1** displays *striated* layers constructed from polar strands of hydrogen-bonded $\text{HN}(\text{SO}_2)_2$ fragments, and apolar strands, in which $(\text{phenyl})_4$ supercycles based upon $\text{C-H}\cdots\text{C}(\pi)$ interactions are fused into a linear superpolymer by $\pi\cdots\pi$ stacking-dimer motifs; moreover, $\text{C-H}\cdots\text{O}$ hydrogen bonds occur between adjacent intra-layer strands and between adjacent layers. Molecule **1** features conformational anomalies induced by the high geometric demands of the supramolecular phenyl pattern. In contrast to **1**, compound **2** forms *lamellar* layers comprising an internal polar lamella of $\text{N-H}\cdots\text{O}$ connected $\text{HN}(\text{SO}_2)_2$ fragments as well as inductively activated methyl groups, and apolar peripheral regions of phenyl rings. The methyl groups are engaged with all their hydrogen atoms in a two-dimensional $\text{C-H}\cdots\text{O}$ network, whereas the protruding phenyl rings of adjacent layers are interlocked in such a way that short $\text{C-H}\cdots\text{O}$ inter-layer contacts are created. On account of the reduced phenyl population as compared to **1**, $\text{C-H}\cdots\text{C}(\pi)$ and $\pi\cdots\pi$ stacking interactions are absent from the crystal structure of **2**. Structure **3** finally consists of *homogeneous* layers, in which the $\text{N-H}\cdots\text{O}$ based catemers are both re-enforced and cross-linked by $\text{C-H}\cdots\text{O}$ bonds originating from the four activated methylene hydrogen atoms, whereas adjacent layers are separated by methyl-oxygen contacts with $\text{H}\cdots\text{O}$ distances longer than the van der Waals limit.

Key words: $\text{N-H}\cdots\text{O}/\text{C-H}\cdots\text{O}$ Hydrogen Bonding, $\text{C-H}\cdots\text{C}(\pi)$ Interactions, $\pi\cdots\pi$ Stacking, Layered Compounds, Sulfonamides