

# 7-*O*-Methylcyanidin 3-*O*- $\beta$ -D-Galactopyranoside, a Novel Anthocyanin from Mango (*Mangifera indica* L. cv. 'Tommy Atkins') Peels

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Z. Naturforsch. **60b**, 801 – 804 (2005); received February 7, 2005

Mango (*Mangifera indica* L.) is one of the most important tropical fruits and its pulp is known to be a rich source of carotenoids. Investigations on the phenolic composition of the peels of a red-colored cultivar (cv. 'Tommy Atkins') revealed the presence of a new anthocyanin, 7-*O*-methylcyanidin 3-*O*- $\beta$ -D-galactopyranoside. Its structure was elucidated by extensive 1D and 2D NMR studies, MS, and chemical transformation.

**Key words:** *Mangifera indica* (L.) cv. 'Tommy Atkins', Anthocyanin, 7-*O*-Methylcyanidin 3-*O*- $\beta$ -D-Galactopyranoside, ESI(+)-MS<sup>2</sup>, NMR

## Introduction

With a global production exceeding 25 million tons in 2003, mangos (*Mangifera indica* L., Anacardiaceae) represent, beside bananas, one of the most important tropical fruits [1]. Mangos are a rich source of carotenoids, with  $\beta$ -carotene accounting for more than half of the total carotenoid content in most cultivars. In tropical countries mangos substantially contribute to the  $\beta$ -carotene supply. While the mango carotenoids have been investigated extensively [2], studies on the phenolic compounds, in particular on the anthocyanins in the peels of red-colored cultivars, are scarce. Recent investigations have shown that the peels of mango fruit contain a number of flavonol and xanthone glycosides as well as gallotannins and benzophenone derivatives [3, 4]. During our studies on the profile of mango peel phenolics, two anthocyanins were detected in red-colored cultivars by HPLC-DAD [5]. The minor anthocyanin could be identified as cyanidin 3-*O*-galactoside by comparison of the HPLC retention time and LC-MS spectrum with reference material. The molecular mass of 463 Da of the predominant anthocyanin suggested the presence of peonidin 3-*O*-galactoside, the only mango peel anthocyanin described in literature so far [6]. However, HPLC experiments with reference compounds could neither confirm the presence of peonidin 3-*O*-galactoside nor of peonidin 3-*O*-glucoside.

Therefore, the objective of the present study was the isolation and structure elucidation of the unknown anthocyanin. It could be unambiguously identified as 7-*O*-methylcyanidin 3-*O*- $\beta$ -D-galactopyranoside (**1**). To the best of our knowledge, this is the first report on 7-*O*-methylcyanidin 3-*O*- $\beta$ -D-galactopyranoside. Furthermore, the anthocyanidin 7-*O*-methylcyanidin has been characterized for the first time using modern analytical methods such as ESI(+)-MS<sup>2</sup>, HRFABMS and NMR.

## Result and Discussion

Mango peels (cv. 'Tommy Atkins') were lyophilized, ground and extracted with aqueous acetone. The extract was purified using a C<sub>18</sub>-Sep-Pak cartridge. Final clean-up of the anthocyanin was achieved by semipreparative reversed-phase HPLC. 7-*O*-Methylcyanidin 3-*O*- $\beta$ -D-galactopyranoside (**1**) was obtained as bluish-red amorphous powder,  $[\alpha]_D^{25} - 237$  (*c* 0.11, 0.1% TFA). The UV/vis spectrum in MeOH, acidified with 0.1% HCl, showed a maximum at 523 nm typical of anthocyanins. The HRFABMS measurements exhibited a [M]<sup>+</sup> ion of *m/z* 463.1203, corresponding to the molecular formula C<sub>22</sub>H<sub>23</sub>O<sub>11</sub>. Collision-induced fragmentation during the ESI(+)-MS<sup>2</sup> of the anthocyanin yielded the predominant fragment with *m/z* 301. The loss of 162 amu indicated

the presence of a hexose moiety. The observation of an anomeric proton at  $\delta = 5.00$  (d,  $J = 7.6$  Hz,  $1''$ -H) together with six protons in the region  $\delta = 3.70$ – $3.91$  confirmed the presence of a sugar moiety in **1**. The connectivities of the seven sugar protons were determined by COSY and 1D-TOCSY, and their directly bonded carbons were assigned by GHSQC. The observed coupling constant of  $J = 7.6$  Hz for the anomeric proton suggested a  $\beta$ -linkage of the sugar moiety to the aglycone. A vicinal coupling constant of  $J = 9.6$  Hz between  $2''$ -H ( $\delta = 3.77$ ) and  $3''$ -H ( $\delta = 3.70$ ),  $J = 3.9$  Hz between  $3''$ -H and  $4''$ -H ( $\delta = 3.91$ , bd) along with NOE's between  $1''$ -H and  $3''$ -H as well as  $1''$ -H and  $5''$ -H established a galactopyranosyl moiety with  ${}^4C_1$  conformation in the case of a D configured pyranose. Further analysis of the  ${}^1H$  NMR, the COSY spectrum and 1D TOCSY spectra revealed an aromatic ABM splitting system consisting of the protons at  $\delta = 6.91$  (bs,  $2'$ -H),  $\delta = 6.20$  (d,  $J = 7.8$  Hz,  $5'$ -H) and  $\delta = 7.22$  (bd,  $J = 7.8$  Hz,  $6'$ -H), a set of two *meta*-coupled protons at  $\delta = 6.13$  (bs,  $6$ -H) and  $\delta = 6.19$  (bs,  $8$ -H), a lowfield-shifted proton at  $\delta = 8.16$  (bs,  $4$ -H), and a methoxy group at  $\delta = 3.63$  (s). The  ${}^{13}C$  chemical shifts of their corresponding carbons C- $2'$  ( $\delta = 118.1$ ), C- $5'$  ( $\delta = 118.3$ ), C- $6'$  ( $\delta = 128.7$ ), C- $6$  ( $\delta = 103.4$ ), C- $8$  ( $\delta = 94.0$ ), C- $4$  ( $\delta = 134.5$ ) were assigned by the GHSQC experiment. By combining all the  ${}^1H$ - ${}^{13}C$  long range correlations (HMBC) of the aromatic protons and the remaining nine quaternary aromatic carbons, the anthocyanidin structure for the aglycone was established. HMBC-correlations between the methoxy group and the quaternary carbon at  $\delta = 170.6$  (C- $7$ ) along with  ${}^1H$ - ${}^{13}C$  long range correlations between both  $6$ -H and C- $7$  and  $8$ -H and C- $7$  indicated that the methoxy group is attached to the A-ring. ROESY crosspeaks between the methoxy group and  $6$ -H as well as  $8$ -H unambiguously established the C- $7$  position of the methoxy group. Furthermore, the HMBC correlation between the anomeric proton  $1''$ -H and C- $4$  at  $\delta = 134.5$  proved the glycosidic linkage of the galactopyranosyl moiety to C- $4$  of the aglycone. Acid hydrolysis (2 N HCl) of **1** yielded the aglycone, 7-*O*-methylcyanidin (**2**), and the sugar moiety, which was identified as D-galactose by  ${}^1H$  NMR and optical rotation ( $[\alpha]_D^{25} = +85$ ) in comparison with an authentic sample. The structure of 7-*O*-methylcyanidin (**2**) was confirmed by MS and extensive 1D and 2D NMR including GCOSY, ROESY, GHSQC and GHMQC. In the ROESY spectrum of **2** crosspeaks between the two aromatic B-ring

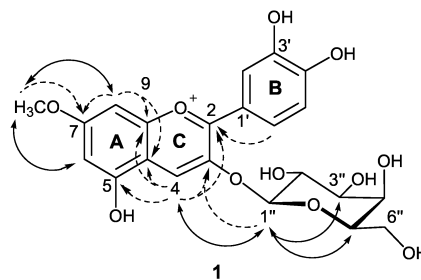


Fig. 1. Structure of 7-*O*-methylcyanidin 3-*O*- $\beta$ -D-galactopyranoside (**1**). Arrows indicate important HMBC [--->] and ROESY [—>] correlations.

protons  $2'$ -H/ $6'$ -H and the aromatic A-ring proton at  $\delta = 7.15$  ( $\delta_C = 92.3$ ) proved its 8-position and thus the 6-position of its *meta*-coupled partner proton at  $\delta = 6.68$  ( $\delta_C = 102.9$ ). Despite the fact that NMR data of the aglycone **2** are not available in the literature, the observed NMR data (see experimental) are in good agreement with the NMR data of similar compounds like cyanidin and cyanidin glycosides [7–9].

In a previous study, an anthocyanin from the flowers of *Bombax malabaricum* DC (Bombacaceae) was isolated and its structure assigned to 7-*O*-methylcyanidin 3-*O*-glucoside by paper-chromatography, IR- and UV/vis spectroscopy, and chemical reactions [10]. Hitherto, 7-*O*-methylcyanidin is not considered in reviews concerning anthocyanidins [11, 12], and, to the best of our knowledge, the present work marks the first report on 7-*O*-methylcyanidin using modern analytical methods such as ESI(+)-MS and NMR.

## Experimental Section

### General experimental procedures

Optical rotation was measured on a Perkin-Elmer model 341 polarimeter with 0.1% aqueous TFA as the solvent. UV spectra were recorded in MeOH, acidified with 0.1% HCl, on a Perkin-Elmer Lambda 20 instrument. IR spectra were obtained using a Perkin-Elmer FT-IR spectrometer. ESI(+)-MS analyses were performed using a Bruker Esquire 3000+ ion trap mass spectrometer fitted with an ESI source.  ${}^1H$  NMR data were obtained on a Varian Unity Inova 500 MHz spectrometer. The  ${}^1H$  and  ${}^{13}C$  chemical shifts were referenced to solvent signals at  $\delta_{H/C} 3.32/49.0$  (methanol- $d_4$  containing 5% TFA) and  $\delta_{H/C} 4.70$  (H $_2$ O)/164.5 (TFA) (D $_2$ O containing 1% TFA) relative to TMS. All 1D ( ${}^1H$ , DPGNOE, TOCSY) and 2D (GCOSY, ROESY, GHSQC, GHMQC, G = “gradient enhanced”) NMR measurements were performed using standard Varian pulse sequences.  ${}^1H$ - ${}^{13}C$  correlation spectra were recorded by GHSQC ( $J_{C-H} = 140$  Hz) for the determination of proton-bearing carbons and

GHMQC ( $^nJ_{C-H}$  optimized for 8 Hz) for multibond correlations (HMBC). ROESY experiments were carried out with 0.5 s mixing time.  $^{13}C$  NMR data were obtained on a Varian Unity Inova 300 MHz spectrometer.

#### Plant material

Mature Peruvian mango fruits (cv. 'Tommy Atkins') obtained from the local market were washed and the peels were removed with a stainless steel knife. After the separation of residual pulp from the peel with a razor blade, the samples were immediately lyophilized and finely ground.

#### Extraction and isolation

Lyophilized peels (125 g) were extracted with aqueous acetone (80 %, v/v; 1% ascorbic acid). The organic solvent was evaporated from the extract *in vacuo* at 30 °C. Aliquots of 30 ml of the aqueous extract were applied on a 10 g C<sub>18</sub>-Sep-Pak cartridge. Non-anthocyanin substances were subsequently eluted with 0.01% HCl and EtOAc. The anthocyanins were eluted with acidified MeOH (0.01% hydrochloric acid) and the eluate was evaporated *in vacuo* to dryness. Using a Bischoff Nitrile column (250 × 15 mm i.d., particle size 5  $\mu$ m), the anthocyanin was purified by semipreparative HPLC. The mobile phase consisted of 87% (v/v) water, 0.1% TFA, 3% acetonitrile (eluent A) and 50% water, 0.1% TFA, 50% acetonitrile (eluent B). The gradient program was as follows: 0% B (5 min), 0% B to 15% B (15 min), 15% B to 100% B (10 min), 100% B (2.5 min), 100% B to 0% B (2.5 min). Monitoring was performed at 370 nm at a flow rate of 6 ml/min. The eluates were evaporated *in vacuo* to dryness, and the residue (16.1 mg) was stored at –80 °C in a nitrogen atmosphere.

#### Acid hydrolysis

The purified sample (1.6 mg) was hydrolyzed under reflux in a nitrogen atmosphere (100 °C, 120 min) using 10 ml of 2 N hydrochloric acid. After cooling, the solution was applied on an activated 1000 mg C<sub>18</sub>-Sep-Pak cartridge and the latter was rinsed with 25 ml water. The carbohydrate containing effluent was collected and final purification was performed by HPLC (Phenomenex, Rezex RCM Monosaccharide, 300 × 7.8 mm; water as mobile phase, 0.55 ml/min, 80 °C). D-galactose (0.4 mg) was identified by  $^1H$  NMR and optical rotation ( $[\alpha]_D^{25} + 85$ ) in comparison with an authentic

sample ( $[\alpha]_D^{25} + 83$ ) obtained from Sigma. The anthocyanidin **2** was eluted from the C<sub>18</sub>-Sep-Pak cartridge with acidified MeOH (0.01% hydrochloric acid) and the eluate was evaporated to dryness *in vacuo*.

#### 7-*O*-Methylcyanidin 3-*O*- $\beta$ -D-galactopyranoside (**1**)

Bluish-red amorphous powder. –  $[\alpha]_D^{25} - 237$  ( $c = 0.11$ , 0.1% TFA). – UV/vis (MeOH, 0.1% HCl):  $\lambda_{max}(lg\epsilon) = 282$  nm (3.98), 523 nm (4.13). – IR (ATR):  $\tilde{\nu} = 3109$ , 1673, 1327, 1280, 1181, 1129  $cm^{-1}$ . –  $^1H$  NMR (500.14 MHz, D<sub>2</sub>O, 1% TFA):  $\delta = 3.63$  (s, 3 H, OMe), 3.70 (dd,  $J = 9.6$ , 4.7 Hz, 1 H, 3''-H), 3.70 (m, 2 H, 6''-H), 3.77 (dd,  $J = 9.6$ , 7.9 Hz, 1 H, 2''-H), 3.77 (bt,  $J = 8.5$  Hz, 1 H, 5''-H), 3.91 (bd,  $J = 3.9$  Hz, 1 H, 4''-H), 5.00 (d,  $J = 7.6$  Hz, 1 H, 5-H), 6.13 (bs, 1 H, 6-H), 6.19 (bs, 1 H, 8-H), 6.20 (d,  $J = 7.8$  Hz, 1 H, 5'-H), 6.91 (bs, 1 H, 2'-H), 7.22 (bd,  $J = 7.9$  Hz, 1 H, 6'-H), 8.16 (bs, 1 H, 4-H). –  $^{13}C$  { $^1H$ } NMR (75.42 MHz, D<sub>2</sub>O, 1% TFA):  $\delta = 58.7$  (OMe), 62.7 (C-6''), 70.1 (C-4''), 72.0 (C-2''), 74.7 (C-3''), 77.9 (C-5''), 94.0 (C-8), 103.4 (C-1''), 103.4 (C-6), 113.1 (C-10), 118.1 (C-2'), 118.3 (C-5'), 120.1 (C-1'), 128.7 (C-6'), 134.5 (C-4), 145.5 (C-3), 146.3 (C-3'), 155.5 (C-4'), 155.8 (C-9), 157.5 (C-5), 161.2 (C-2), 170.6 (C-7). – ESI(+)-MS<sup>2</sup>:  $m/z$  463 [ $M^+$ ], 301 [ $M^+$ -galactopyranosyl+H]. – HRFABMS:  $m/z$  463.1203 (calcd. for C<sub>22</sub>H<sub>23</sub>O<sub>11</sub>: 463.1239).

#### 7-*O*-Methylcyanidin (**2**)

$^1H$  NMR (500.14 MHz, CD<sub>3</sub>OD, 5% TFA):  $\delta = 4.03$  (s, 3 H, OMe), 6.68 (d,  $J = 2.1$ , 1 H, 6-H), 7.04 (d,  $J = 8.7$  Hz, 1 H, 5'-H), 7.15 (dd,  $J = 0.8$ , 2.0 Hz, 1 H, 8-H), 8.19 (d,  $J = 2.3$  Hz, 1 H, 2'-H), 8.31 (dd,  $J = 2.2$ , 8.7 Hz, 1 H, 6'-H), 8.61 (d,  $J = 0.7$  Hz, 1 H, 4-H). –  $^{13}C$  NMR (indirectly determined by GHSQC and GHMQC at 500.14 MHz):  $\delta = 57.2$  (OMe), 92.3 (C-8), 102.9 (C-6), 114.0 (C-10), 117.3 (C-5'), 118.2 (C-2'), 121.8 (C-1'), 127.6 (C-6'), 133.3 (C-4), 147.4 (C-3'), 147.7 (C-3), 155.7 (C-4'), 156.6 (C-9), 157.2 (C-5), 163.1 (C-2), 169.2 (C-7). – ESI(+)-MS<sup>2</sup>:  $m/z$  301 [ $M^+$ ].

#### Acknowledgements

The authors thank Mr. Norbert Nieth, Institute of Organic Chemistry, University of Heidelberg, for performing HRFABMS analysis and Mrs. Elena Merisor, Institute of Chemistry, Hohenheim University for recording of the IR spectrum. Financial support by fruit – International Fruit Foundation, Heidelberg-Schlierbach, Germany, is gratefully acknowledged.

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