## Heterobimetallische 3d-4f-Komplexe mit Bis(1,2-dithiooxalato)nickelat(II) als planarem Brückenbaustein

Heterobimetallic 3d-4f-Complexes with Bis(1,2-dithiooxalato)nickelate(II) as Planar Bridging Block

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Planar bis(1,2-dithiooxalato)nickelates(II) react in aqueous solutions of lanthanide ions to form pentanuclear, heterobimetallic complexes of the general composition  $[\{Ln(H_2O)_n\}_{2-}\{Ni(dto)_2\}_{3}]\cdot xH_2O$  (Ln = Y<sup>3+</sup>, La<sup>3+</sup>, Ce<sup>3+</sup>, Pr<sup>3+</sup>, Nd<sup>3+</sup>, Sm<sup>3+</sup>, Eu<sup>3+</sup>, Gd<sup>3+</sup>, Tb<sup>3+</sup>, Dy<sup>3+</sup>, Ho<sup>3+</sup>, Er<sup>3+</sup>, Tm<sup>3+</sup>, Yb<sup>3+</sup>, Lu<sup>3+</sup>; n=4 or 5; x=9-12). With  $[\{Nd(H_2O)_5\}_{2}\{Ni(S_2C_2O_2)_2\}_{3}]\cdot xH_2O$  (x=10-12) (1) and  $[\{Er(H_2O)_4\}_{2}\{Ni(S_2C_2O_2)_2\}_{3}]\cdot xH_2O$  (x=9-10) (2) we were able to isolate two complexes of this series as single crystals, which were characterized by X-ray structure analysis. Depending on the individual ionic radii of the lanthanide ions, the compounds crystallize in two different crystal systems with the following unit cell parameters: 1, monoclinic in  $P2_1/c$  with a=11.3987(13), b=11.4878(8), c=20.823(2) Å,  $\beta=98.907(9)^\circ$  and Z=2; 2, triclinic in  $P\overline{1}$  with a=10.5091(6), b=11.0604(6), c=11.2823(6) Å,  $\alpha=107.899(4)^\circ$ ,  $\beta=91.436(4)^\circ$ ,  $\gamma=112.918(4)^\circ$  and Z=1. The channels and cavities appearing in the packing of the molecules are occupied by uncoordinated water molecules. High magnetic moments up to 14.65 BM./f.u. have been observed at room temperature due to the combined moments of the individual lanthanide ions.

Key words: Lanthanides, 1,2-Dithiooxalate, Crystal Structure, Nickel(II)