

# Preparation, Crystal Structure, Properties, and Electronic Band Structure of $\text{TiTaSe}_3$

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$\text{TiTaSe}_3$  was prepared from a fused mixture of  $\text{Ti}_4\text{Ta}_2\text{Se}_{11}$ , Ta and Se in the molar ratio 1:2:1. The compound shows dimorphism with H- $\text{TiTaSe}_3$ , hexagonal, space group  $P6_3/mmc$ ,  $a = 7.2436(6)$ ,  $c = 5.9736(6)$  Å,  $c/a = 1.213$  and O- $\text{TiTaSe}_3$ , orthorhombic, space group  $Pnma$ ,  $a = 9.554(13)$ ,  $b = 3.6244(6)$ ,  $c = 14.7271(17)$  Å. The crystal structure of H- $\text{TiTaSe}_3$  is isotypic to  $\text{BaVSe}_3$  whereas that of O- $\text{TiTaSe}_3$  is closely related to the  $\text{NH}_4\text{CdCl}_3$ -type. Characteristic features of the structures are:  ${}^1_\infty[\text{TaSe}_3^{2-}]$  chains of regular octahedra sharing faces along [001] for the hexagonal form and columns of double edge-sharing octahedra  ${}^1_\infty[\text{Ta}_2\text{Se}_6^{2-}]$  running along [010] for O- $\text{TiTaSe}_3$ . The columns are each separated by  $\text{Ti}^+$  ions with the coordination number CN = 12 and CN = 8 respectively. The structures are compared and discussed in context with other isotypic structures of chalcogenides. The orthorhombic modification O- $\text{TiTaSe}_3$  is a semiconductor while H- $\text{TiTaSe}_3$  shows conventional metallic behaviour. The electronic structures of both modifications are discussed on the base of band structure calculations performed within the framework of density functional theory.

*Key words:* Thallium Tantalum Selenide, Crystal Structure, Dimorphic, Properties, Electronic Band Structure