## Hydrogen-Bonded Networks Featuring Yttrium(III) Complexes of N,N'-Dimethylurea (DMU): Preparation and Characterization of $[Y(DMU)_6][YCl_6]$ and $[Y(NO_3)_3(DMU)_3]$

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In order to examine the possibility of using yttrium(III) in the crystal engineering of hydrogen-bonded coordination complexes and to compare the molecular and supramolecular  $Y^{III}/Cl^-$  or  $NO_3^-/DMU$  chemistry with the already well-developed  $Ln^{III}/Cl^-$  or  $NO_3^-/DMU$  chemistry ( $Ln^{III}$  = lanthanide, DMU = N,N'-dimethylurea), compounds [Y(DMU)<sub>6</sub>][YCl<sub>6</sub>] (1) and [Y(NO<sub>3</sub>)<sub>3</sub>(DMU)<sub>3</sub>] (2) have been prepared. The structures of both compounds have been determined by single-crystal X-ray diffraction. The structure of 1 consists of octahedral [Y(DMU)<sub>6</sub>]<sup>3+</sup> and [YCl<sub>6</sub>]<sup>3-</sup> ions. The Y<sup>III</sup> ion in 2 is nine-coordinate and ligation is provided by three O-bonded DMU ligands and three bidentate chelating nitrato groups; the coordination polyhedron about the metal can be viewed as a distorted, monocapped square antiprism. The [Y(DMU)<sub>6</sub>]<sup>3+</sup> cations and [YCl<sub>6</sub>]<sup>3-</sup> anions self-assemble to form a hydrogen-bonded 3D architecture in 1. Most of the hydrogen-bonding functionalities on the components of 2 create also a 3D network. Two motifs of interionic/intramolecular hydrogen-bonds have been observed: N-H···Cl in 1 and N-H···O(NO<sub>3</sub><sup>-</sup>) in 2. The IR data are discussed in terms of the nature of bonding and the structures of the two complexes.

Key words: Crystal Structures, Hydrogen-Bonded Coordination Complexes, Vibrational Spectroscopy, Yttrium(III)/N,N'-Dimethylurea Complexes, Yttrium(III) Nitrate Complexes