

Organoaluminium and -gallium Lewis-Acid Adducts of Tetramethylmethylenediamine

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Dedicated to Professor Hubert Schmidbaur on the occasion of his 70th birthday

The compounds $\text{Me}_3\text{Al}-\text{Me}_2\text{NCH}_2\text{NMe}_2-\text{AlMe}_3$ (**1**) and $\text{Me}_3\text{Ga}-\text{Me}_2\text{NCH}_2\text{NMe}_2-\text{GaMe}_3$ (**2**) were prepared by reacting $\text{Me}_2\text{NCH}_2\text{NMe}_2$ (TMMDA) with two equivalents of the metal trialkyls in hydrocarbon solutions. With the ether adduct $\text{Me}_3\text{Ga}\cdot\text{OEt}_2$ $\text{Me}_2\text{NCH}_2\text{NMe}_2$ reacts to give the mono-adduct $\text{Me}_2\text{NCH}_2\text{NMe}_2-\text{GaMe}_3$ (**3**). These compounds were characterized by NMR spectroscopy (^1H , ^{13}C and ^{27}Al) and by elemental analyses. Crystal structure investigations show **1** and **2** to be monomeric and to adopt a *trans,trans*-conformation for their M-N-C-N-M backbones. **3** is also monomeric in the solid state, but adopts a *cis,trans*-conformation. Tetramethylformamidine chloride and also chlorotetramethylformamidine chloride reacts with lithium aluminium hydride to give the mono-adduct $[\text{Me}_2\text{NCH}_2\text{NMe}_2-\text{AlH}_3]_2$ (**4**), which is dimeric and can be regarded as a double TMMDA adduct to Al_2H_6 with five-coordinate Al atoms. *Ab initio* calculations on the MP2/6-311G** level of theory have been performed for the model compound $\text{H}_3\text{N}-\text{H}_2\text{Al}(\mu-\text{H})_2\text{AlH}_2-\text{NH}_3$ to obtain its molecular structure and vibrational spectrum for comparison with **4** and for the assignment of its vibrational spectrum.

Key words: Aluminium, Gallium, Amino Adduct, Crystal Structure, *ab initio* Calculations