Formamidinium Salts of Low Valent Metal Halide Anions MX_3^- (M = Ge, Sn) and $M_2X_6^{2-}$ (M = Ga, In)

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Z. Naturforsch. **59b**, 1524 – 1531 (2004); received August 31, 2004

Dedicated to Professor Hubert Schmidbaur on the occasion of his 70th birthday

Tetramethylformamidinium trichlorogermanite, [CH(NMe₂)₂][GeCl₃], tetramethylformamidinium trichlorostannite, [CH(NMe₂)₂][SnCl₃], *bis*-(tetramethylformamidinium hexaiododigallate, [CH(NMe₂)₂]₂[Ga₂I₆] and *bis*-(tetramethylformamidinium hexachlorodiindate, [CH(N Me₂)₂]₂[In₂Cl₆] have been prepared by the reactions between tetramethylformamidinium chloride, [CH(NMe₂)₂]₂[Cl, and the corresponding low valent halides GeCl₂ (as dioxane adduct), SnCl₂, "GaI" and InCl. Their crystal structures have been determined by single crystal X-ray diffraction. [CH(NMe₂)₂][GeCl₃] aggregates in a centrosymmetric dimeric structure, in which two trigonal pyramidal GeCl₃ units are connected together by two weak Ge···Cl bonds and each Ge atom is bonded to one cation by a weak Ge···N contact. Two sets of weak hydrogen bonds C-H···Cl are observed with bond lengths of 2.87(2) Å and 2.85(2) Å. In [CH(NMe₂)₂][SnCl₃], the SnCl₃ units adopts a (3+3) coordination with three normal Sn-Cl bonds and three weak Sn···Cl contacts. [CH(NMe₂)₂]₂[Ga₂I₆] and [CH(NMe₂)₂]₂[In₂Cl₆] contain metal-metal bonded anions with distorted staggered ethane-like conformations. The metal-metal bond lengths are 2.423(1) Å (Ga-Ga) and 2.719(1) Å (In-In). Their Raman spectra contain intense bands at 118.7 cm⁻¹ (Ga-Ga) and 174.7 cm⁻¹ (In-In) associated with metal-metal stretching modes.

Key words: Formamidinium Salts, Germanite, Stannite, Digallate, Diindate