High-Temperature Synthesis, Crystal Structure, and Properties of the New Sodium Rare-Earth Oxide Borates $Na_2RE_2(BO_3)_2O$ (*RE* = Dy, Ho)

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Z. Naturforsch. 59b, 1200-1208 (2004); received July 20, 2004

Dedicated to Professor Hubert Schmidbaur on the occasion of his 70th birthday

The new monoclinic oxide borates Na₂*RE*₂(BO₃)₂O (*RE* = Dy, Ho) were synthesized using standard solid-state reactions in the temperature range 900–950 °C. They are isotypic to the known phases Na₂*RE*₂(BO₃)₂O (*RE* = Y, La, Nd, Sm-Gd, Er). The single crystal X-ray structure determination of Na₂Dy₂(BO₃)₂O revealed: *P*2₁/*c*, *a* = 1063.9(1), *b* = 626.2(1), *c* = 1025.3(1) pm, β = 117.76(1)°, *Z* = 4, *R*1 = 0.0221, *wR*2 = 0.0402 (all data). The corresponding lattice parameters of Na₂Ho₂(BO₃)₂O determined from powder data are *a* = 1061.2(5), *b* = 623.7(2), *c* = 1022.5(3) pm, and β = 117.7(1)°. The structure consists of infinite sheets of *REO*₈-polyhedra in the *bc*-plane, which are seperated by sodium atoms. The BO₃-groups are isolated forming layers in the *bc*-plane. The results of IR-spectroscopic investigations, temperature-resolved in-situ powder-diffraction measurements, and DTA/TG measurements on Na₂Dy₂(BO₃)₂O are also presented.

Key words: Solid-State Synthesis, Oxide Borates, Crystal Structure