

Intermolecular Interactions in the Crystal of a New Nickel(II)-cobalt(II)-nickel(II) Trinuclear Complex Containing a Macrocyclic Complex Ligand

Xiao-Zeng Li^{a,b}, Bao-Lin Liu^a, Juen-Hong He^a, and Dai-Zheng Liao^b

^a Department of Chemistry, Tianjin University, Tianjin 300072, P. R. China

^b Department of Chemistry, Nankai University, Tianjin 300071, P. R. China

Reprint requests to Dr. X.-Z. Li. Fax: +86-022-27403475. E-mail: tianuleexz@tom.com

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A new trinuclear complex $[\text{Co}(\text{NiL})_2(\text{H}_2\text{O})_2](\text{ClO}_4)_2$ was synthesized by using a macrocyclic complex ligand $[\text{NiL}]$, where L is the dianion of diethyl-5,6,7,8,16,17-hexahydro-6,7-dioxo-16*H*-dibenzo[*e,n*] [1,4,8,12]tetraazacyclopentadecine-13,19-dicarboxylate. X-ray analysis revealed that the two Ni(II) ions have the same distorted N_4 square planar coordination geometries. The Co(II) ion resides in a distorted octahedral O_6 coordination environment. In the crystal, $\pi \cdots \pi$ interaction between a phenyl ring and three non-aromatic π -systems involving Ni and N atoms was observed together with $\text{C}-\text{H} \cdots \text{O}$ and $\text{O}-\text{H} \cdots \text{O}$ hydrogen bonds. These non-covalent interactions link the dications and the perchlorate anions to form a 3D supramolecular network.

Key words: Nickel(II)-Cobalt(II)-Nickel(II) Complex, Crystal Structure, Non-Covalent Interaction, Macrocyclic Oxamido Ligand