

Dimethyl-, Disilyl- and Digermylsulfide: Different Intermolecular Contacts in the Solid State

Norbert W. Mitzel and Udo Losehand

Westfälische Wilhelms-Universität Münster, Institut für Anorganische und Analytische Chemie,
Corrensstr. 30, D-48149 Münster, Germany

Reprint requests to Prof. Dr. N.W. Mitzel. Fax (+49)251 83 36007. E-mail: Mitzel@uni-muenster.de

Z. Naturforsch. **59b**, 635 – 638 (2004); received March 22, 2004

The compounds $(\text{H}_3\text{C})_2\text{S}$, $(\text{H}_3\text{Si})_2\text{S}$ and $(\text{H}_3\text{Ge})_2\text{S}$ have been crystallised *in situ* on a diffractometer and their crystal structures determined by low-temperature X-ray diffraction. The molecules are present as monomers in the crystals. The aggregation of the molecules through secondary intermolecular contacts in the crystal is different: $(\text{H}_3\text{C})_2\text{S}$ is weakly associated into dimers by $\text{S}\cdots\text{S}$ contacts, whereas $(\text{H}_3\text{Si})_2\text{S}$ and $(\text{H}_3\text{Ge})_2\text{S}$ form $\text{Si}\cdots\text{S}$ and $\text{Ge}\cdots\text{S}$ contacts in an ice-analogous aggregation motif. Important geometry parameters are $(\text{H}_3\text{C})_2\text{S}$: C-S 1.794(av) Å, C-S-C 99.2(1)°; $(\text{H}_3\text{Si})_2\text{S}$: Si-S 2.143(1) Å, Si-S-Si 98.4°; $(\text{H}_3\text{Ge})_2\text{S}$ Ge-S 2.223(2) and 2.230(2) Å, Ge-S-Ge 98.2(1)°.

Key words: Silicon, Germanium, Sulphur, Crystal Structure, Secondary Bonds