

Magnetic Properties and Crystal Structure of a Cu^{II}Gd^{III} Heterodinuclear Schiff Base Complex

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The crystal structure and the magnetic properties of a heterodinuclear complex, [LCu(Me₂CO)Gd(NO₃)₃]₂ (L=N,N'-bis(2-hydroxy-3-methoxybenzylidene)-1,3-diaminopropane) are reported: [(C₁₉H₂₀N₂O₄)Cu(C₃H₆O)Gd(NO₃)₃]₂, triclinic, space group *P*1, with *a* = 12.118.3(9), *b* = 13.562(3), *c* = 9.391(3) Å, α = 93.03(3), β = 107.65(2), γ = 73.07(2)°, *V* = 1406.0(7) Å³, *Z* = 1. The crystal structure consists of two independent binuclear Cu^{II}Gd^{III} complexes and two non-coordinating acetone molecules in the asymmetric unit. The central region of the complex is occupied by Cu^{II} and Gd^{III} ions which are bridged by two phenolato oxygen atoms of the ligand. The Cu^{II} ion is in a square-planar geometry and coordinated by four donor atoms of the ligand (N₂O₂). The Gd^{III} ion is deca-coordinated. In addition to the two phenolate oxygen atoms, its coordination sphere contains two oxygen atoms of the OMe side arms of L and six oxygen atoms from the three bidentate nitrate ions. The average Cu...Gd separation is 3.375(2) Å. The χT versus *T* plots, χ being the molar magnetic susceptibility per Cu^{II}Gd^{III} unit and *T* the temperature, has been measured in the 4 – 347 K range. The values of the intrachain interaction parameters have been deduced from the magnetic data: *J* = 7.4 cm⁻¹, *g*_{Cu} = 2.12, *g*_{Gd} = 2.06. This indicates a weak ferromagnetic spin exchange interaction between Cu^{II} and Gd^{III} ions. The nature of the magnetic super-exchange interaction of the title compound is compared with similar Cu^{II}Gd^{III} heterodinuclear complexes.

Key words: Heterodinuclear Complex, Copper, Gadolinium, Crystal Structure, Magnetic Properties