Synthesis and Structural Relationship of the Ternary Indides $Sc_3Ni_{2.10(5)}In_{3.60(5)}, Sc_3Ni_{2.14(2)}In_{3.76(2)}, ScPd_{0.981(2)}In$ and $Sc_3Rh_{1.594(9)}In_4$

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Dedicated to Professor Ingo-Peter Lorenz on the occasion of his 60th birthday

The ternary scandium transition metal indides $Sc_3Ni_{2.10(5)}In_{3.60(5)}$, $Sc_3Ni_{2.14(2)}In_{3.76(2)}$, $ScPd_{0.981(2)}In$, and $Sc_3Rh_{1.594(9)}In_4$ have been synthesized from the elements in glassy carbon crucibles in a high frequency furnace or by arc-melting. They have been investigated by X-ray powder diffraction and the structures refined on the basis of single crystal diffractometer data: $Lu_3Co_{1.87}In_4$ type, $P\bar{6}$, a=745.7(1), c=342.85(7) pm, wR2=0.0689, 545 F^2 , 24 parameters for $Sc_3Ni_{2.10(5)}In_{3.60(5)}$, a=753.63(7), c=344.3(1) pm, wR2=0.0362, 792 F^2 , 22 parameters for $Sc_3Ni_{2.14(2)}In_{3.76(2)}$, $P\bar{6}2m$, ZrNiAl type, a=764.1(2), c=345.90(8) pm, wR2=0.0333, 326 F^2 , 15 parameters for $Sc_3Ni_{2.14(2)}In_{3.76(2)}$, In, and In, and In, In, In, the three structure types the scandium atoms build trigonal prisms. The latter are filled exclusively by palladium atoms in $ScPd_{0.981(2)}In$, while transition metal and indium atoms fill these sites in the other three structures. The different coloring of the trigonal prismatic sites leads to a symmetry reduction for the structures of $Sc_3Ni_{2.10(5)}In_{3.76(5)}$, $Sc_3Ni_{2.14(2)}In_{3.76(2)}$ and $Sc_3Rh_{1.594(9)}In_4$. The structural relationship is described on the basis of a group-subgroup scheme. Chemical bonding in these intermetallics is briefly discussed.

Key words: Indium, Crystal Structure, Scandium