## Ternary Rare Earth Metal Silicides $RE_2RE'_3Si_4$ with Orthorhombic $Sm_5Ge_4$ or Tetragonal $Zr_5Si_4$ Type Structure

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Dedicated to Professor Ingo-Peter Lorenz on the occasion of his 60<sup>th</sup> birthday

Ternary silicides  $RE_2RE_2'Si_4$  (RE = La, Ce; RE' = Y, Lu) were synthesized by arc-melting of the elements. Single crystals were grown by annealing the arc-molten buttons slightly below the melting points in water-cooled silica tubes in an induction furnace. Five silicides  $RE_2RE_3'Si_4$  were investigated by X-ray powder and single crystal diffraction:  $Zr_5Si_4$  type,  $P4_12_12$ , a = 779.4(3), c = 1441.3(9) pm, wR2 = 0.072, 1806  $F^2$  values, 45 variables for  $La_{1.72(4)}Y_{3.28(4)}Si_4$ ,  $P4_32_12$ , a = 769.92(7), c = 1412.3(1) pm, wR2 = 0.079, 1846  $F^2$  values, 45 variables, BASF = 0.36(4) for  $La_{1.72(2)}Lu_{3.28(2)}Si_4$ ,  $P4_12_12$ , a = 778.6(1), c = 1433.9(3) pm, wR2 = 0.054,  $1910 F^2$  values, 46 variables, BASF = 0.34(4) for  $Ce_{1.82(6)}Lu_{3.18(6)}Si_4$ ,  $P4_32_12$ , a = 778.8(3), c = 1436.0(10) pm, wR2 = 0.166, 1916  $F^2$  values, 45 variables for  $Ce_{1.71(7)}Y_{3.29(7)}Si_4$ , and  $Sm_5Ge_4$  type, Pnma, a = 749.2(2), b = 1484.2(7), c = 780.3(2) pm, wR2 = 0.070, 1956  $F^2$  values and 49 variables for Ce<sub>1.47(3)</sub> Y<sub>3.53(3)</sub>Si<sub>4</sub>. The local coordinations of both structure types are very similar, i. e. CN 18, 16, and 14 coordination polyhedra for the RE1, RE2, and RE3 atoms. The main structural motif is the eightfold coordination of the RE3 positions by rare earth metal atoms. These slightly distorted RERE'<sub>8</sub> cubes are condensed via common corners (Zr<sub>5</sub>Si<sub>4</sub> type) or via common edges (Sm<sub>5</sub>Ge<sub>4</sub> type). All silicon atoms in these silicides form Si<sub>2</sub> pairs at Si-Si distances ranging from 254 to 258 pm. DFT band structure calculations confirm the metallic character of both silicides and the trivalent state of cerium in Ce<sub>5</sub>Si<sub>4</sub>. The COHP analysis (Crystal Orbital Hamilton Population) of the Si-Si bonds shows a significant occupation of Si-Si antibonding orbitals, which is partially compensated by a Si–Si  $\pi *$ ,  $\sigma * \rightarrow Y-4d$  (Ce-5d) back donation.

Key words: Silicide, Crystal Structure, Solid State Synthesis, Chemical Bonding

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