Investigations in the Systems Ag-Hg-X-O ( $X = As^V$ ,  $Se^{IV}$ ,  $Se^{VI}$ ): Hydrothermal Single Crystal Growth of  $Ag_3AsO_4$ ,  $AgHg^I_2AsO_4$ ,  $AgHg^IIAsO_4$ ,  $Ag_2SeO_4$  and the Crystal Structure of  $Ag_2Hg^{II}(SeO_3)_2$ 

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Single crystals of the already known phases Ag<sub>3</sub>AsO<sub>4</sub>, AgHg<sup>I</sup><sub>2</sub>AsO<sub>4</sub>, AgHg<sup>II</sup>AsO<sub>4</sub>, Ag<sub>2</sub>SeO<sub>4</sub> and of the hitherto unknown compound Ag<sub>2</sub>Hg<sup>II</sup>(SeO<sub>3</sub>)<sub>2</sub> were obtained under hydrothermal conditions (250 °C, 5 d) from starting mixtures of the metal nitrates and the respective acids. Both Ag<sub>3</sub>AsO<sub>4</sub> and AgHg<sup>I</sup><sub>2</sub>AsO<sub>4</sub> are isotypic with the corresponding phosphates, Ag<sub>3</sub>PO<sub>4</sub> and AgHg<sup>I</sup><sub>2</sub>PO<sub>4</sub>, whereas AgHg<sup>II</sup>AsO<sub>4</sub> and Ag<sub>2</sub>SeO<sub>4</sub> crystallize in the thenardite (Na<sub>2</sub>SO<sub>4</sub> (V)) structure. All crystal structures were refined by means of single crystal X-ray data. The crystal structure of Ag<sub>2</sub>Hg<sup>II</sup>(SeO<sub>3</sub>)<sub>2</sub> [Pbca, Z = 8, a = 6.8206(11), b = 11.237(3), c = 16.876(2) Å, 1677 structure factors, 101 parameters,  $R[F^2 > 2\sigma(F^2)] = 0.0193$ ,  $wR(F^2 \text{ all}) = 0.0394$ ] consists of considerably distorted [AgO<sub>6</sub>] and  $[HgO_6]$  octahedra, and trigonal Se<sup>IV</sup>O<sub>3</sub> pyramids as the main building units. The  $[MO_6]$  octahedra build a complex framework by sharing common edges and corners, and the Se<sup>IV</sup>O<sub>3</sub> pyramids are located in the vacancies of this arrangement. The average Hg-O distance of 2.399 Å is significantly shorter than the average Ag-O distance of 2.551 Å. The geometries of the two crystallographically independent Se<sup>IV</sup>O<sub>3</sub> pyramids are very similar and the average distance and angle  $(\bar{d}(Se-O) = 1.709 \text{ Å})$ ,  $\angle$ (O-S-O) = 100.1°) lie in the characteristic range for a selenite(IV) group. A short comparative structural discussion between the various compounds obtained during the hydrothermal experiments is given.

*Key words:* Silver, Mercury, Selenites(IV), Selenates(VI), Arsenates(V)