A Novel Methylene Dithioether as a Ligand: Synthesis and Molecular Structure of a Zinc(II) Complex with  $N_4S_2$  Coordination Environment

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Z. Naturforsch. **58b**, 1027 – 1029 (2003); received July 21, 2003

The octadentate ligand [N(CH<sub>2</sub>CH<sub>2</sub>NH<sub>2</sub>)(CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub> OH)(CH<sub>2</sub>CH<sub>2</sub>S)]<sub>2</sub>CH<sub>2</sub>, (NNOS-232)<sub>2</sub>CH<sub>2</sub>, was synthesized accidentally by the reaction of the unsymmetrically substituted tripod [N(CH<sub>2</sub>CH<sub>2</sub>NH<sub>2</sub>)(CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>OH)(CH<sub>2</sub> CH<sub>2</sub>SH)], NNOS-232, with dichloromethane in the presence of aluminum hydroxide. Ligand (NNOS-232)<sub>2</sub>CH<sub>2</sub> was reacted with zinc bis(perchlorate) hexahydrate to yield the complex [Zn((NNOS-232)<sub>2</sub>CH<sub>2</sub>)](ClO<sub>4</sub>)<sub>2</sub> **1** exhibiting a distorted octahedrally coordinated zinc atom in an N<sub>4</sub>S<sub>2</sub> coordination environment, as shown by an X-ray diffraction study.

Key words: Zinc, Dithioacetal, Tripodal Ligands