

# Crystal Structure and Magnetic Properties of a Dinuclear Iron(III) Doubly Oxygen Bridged Schiff Base Complex

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[Fe(L)Cl]<sub>2</sub> (L = N-(4-methylphenyl)-3-methoxy-salicylaldimimine) was synthesized and its crystal structure determined. [C<sub>30</sub>H<sub>26</sub>Cl<sub>2</sub>Fe<sub>2</sub>N<sub>2</sub>O<sub>6</sub>], triclinic, space group *P*  $\bar{1}$ , *a* = 9.278(2), *b* = 9.4050(10), *c* = 10.489(2) Å,  $\alpha$  = 64.43(2),  $\beta$  = 74.540(10),  $\gamma$  = 62.40(2)°, *V* = 729.1(2) Å<sup>3</sup>, *Z* = 1. Two identical [Fe(L)Cl] fragments, related by an inversion center, are connected by two bridging O atoms to form a binuclear unit. The iron(III) centers are separated by 3.196(2) Å and weakly antiferromagnetically coupled (*J* = −10.1(1) cm<sup>−1</sup>), as derived from temperature-dependent magnetic susceptibility measurements in the range 5.1–283.5 K.

*Key words:* Dinuclear Iron(III) Complex, Super-Exchange Interactions, Antiferromagnetic Coupling