Preparation and Crystal Structures of Ternary Rare Earth Silver and Gold Arsenides $LnAgAs_2$ and $LnAuAs_2$ with Ln = La-Nd, Sm, Gd, and Tb

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Dedicated to Professor Mewis on the occasion of his 60th birthday

The 14 arsenides $LnAgAs_2$ and $LnAuAs_2$ (Ln = La-Nd, Sm, Gd, Tb) were prepared by reaction of stoichiometric mixtures of the elemental components at high temperatures and characterized by Xray diffractometry. The silver compounds LaAgAs₂ and CeAgAs₂ and the gold compounds LnAuAs₂ (Ln = Ce-Nd, Sm, Gd, Tb) crystallize with HfCuSi₂ type structure (P4/nmm, Z = 2). Of these, the structures of CeAgAs₂ (a = 408.5(1), c = 1048.2(1) pm, conventional residual R = 0.017 for 261 structure factors and 12 variable parameters) and CeAuAs₂ (a = 411.4(1), c = 1015.3(2) pm, R =0.030 for 428 F values) were refined from four-circle diffractometer data. The silver compounds $LnAgAs_2$ (Ln = Pr, Nd, Sm, Gd, Tb) are isotypic with the antimonide $SrZnSb_2$ (Pnma, Z = 4) as demonstrated by a single-crystal structure refinement of PrAgAs₂ (a = 2107.3(4), b = 401.7(1), c =407.8(1) pm, R = 0.042 for 746 F values and 26 variables). The gold compound LaAuAs₂ (I4/mmm, Z = 4, a = 416.9(1), c = 2059.5(3) pm, R = 0.038 for 303 F values and 13 variables) was found to be isotypic with the bismuthide SrZnBi₂, again by a refinement from single-crystal diffractometer data. In the structures of CeAgAs₂, LaAuAs₂, and CeAuAs₂ large displacement parameters perpendicular to the four-fold axes were found for one of the two arsenic positions. These structures could also be refined with split positions for these arsenic atoms, which allow for considerable As-As bonding, resulting in a formal charge of -1 for these atoms. Chemical bonding in these compounds can thus be rationalized by a simple model corresponding to the formula $Ln^{+3}T^{+1}As^{-1}As^{-3}$ (T = Ag, Au), where the superscripts indicate oxidation numbers.

Key words: Rare Earth Compounds, Arsenides, Crystal Structure