

Preparation and Crystal Structure of Hexapotassium- μ -oxo-hexathiodititanate(IV), $\text{K}_6\text{Ti}_2\text{OS}_6$

Kurt O. Klepp and Ferdinand Fabian

Department of General and Inorganic Chemistry, University of Linz,
Altenbergerstr. 69, A-4040 Linz, Austria

Reprint requests to Prof. K. O. Klepp. E-mail: kurt.klepp@jku.at

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Pale yellow crystals of the title compound were obtained by reacting an intimate mixture of K_2S , $\text{K}_2\text{S}_2\text{O}_4$, Ti and S at 650 °C. $\text{K}_6\text{Ti}_2\text{OS}_6$ is monoclinic, *mP*60, s. g. $P2_1/c$. It is characterized by the formation of bitetrahedral complex anions, $[\text{S}_3\text{TiOTiS}_3]^{6-}$, which adopt a staggered conformation. The mean Ti-S and Ti-O bond lengths are 2.242(1) and 1.836(2) Å, respectively, the Ti-O-Ti bond angle is 153.8(2)°. Two K^+ ions complete the coordination of the bridging oxygen atom by the cations to a heavily distorted tetrahedral configuration. The anions are arranged in corrugated slabs running parallel to (100). The packing and crystallographic relationship to $\text{K}_6\text{Co}_2\text{O}_7$ are discussed.

Key words: Chalcogenides, Titanium, Complex Oxysulfides