

# The Reduction of Pyridine to 1,4-Dihydropyridine by Lithiumtetrahydroborate in the Presence of Water

Heinrich Nöth and Marcus Warchhold

Department of Chemistry, University of Munich, Butenandtstr. 5 – 13, D-81 377 München, Germany

Reprint requests to Prof. Dr. H. Nöth. E-mail: H.Noeth@lrz.uni-muenchen.de

Z. Naturforsch. **58b**, 123 – 126 (2003); received October 11, 2002

In the presence of water lithium tetrahydroborate is capable of reducing pyridine to 1,4-dihydropyridine. The product, lithiumtetrakis(pyridine) tetrakis(1,4-hydropyridyl)borate, was isolated and characterized by NMR and IR spectroscopy and single crystal X-ray diffraction.  $[\text{Li}(\text{py})_4][\text{B}(\text{pyH})_4] \cdot 2(1,4\text{-dioxane})$  crystallized in space group  $P4_2/n$ ,  $Z = 2$ . Similarly,  $\text{NaBH}_4$  can also reduce pyridine in the presence of water and equivalent amounts of  $\text{LiCl}$  or  $\text{ZnCl}_2$ .

*Key words:* Reduction of Pyridine, Tetrakis(hydropyridino)borate, X-Ray Structure