

New Information on the Na-Ti-Se Ternary System

Jennifer A. Aitken and Mercuri G. Kanatzidis

Department of Chemistry and Center for Fundamental Materials Research,
Michigan State University, East Lansing, MI 48824.

Reprint requests to Prof. Dr. Mercuri G. Kanatzidis. E-mail: kanatzid@cem.msu.edu

Z. Naturforsch. **56 b**, 49–56 (2001); received August 22, 2000

Molten Salt Fluxes, Polychalcogenide, Polyselenide

$\text{Na}_2\text{Ti}_2\text{Se}_9$ was first discovered in the quaternary system Na/Ti/P/Se, but it can be prepared rationally by dissolving Ti in a Na_2Se_x flux at 375 °C. The silverish dark red crystals are air- and water-stable. A new method for synthesizing Na_4TiSe_4 is also reported. $\text{Na}_2\text{Ti}_2\text{Se}_9$ crystallizes in space group $P2_1$ with $a = 12.785(4)$, $b = 14.848(4)$, $c = 13.961(4)$ Å, $\beta = 94.556(4)^\circ$, $R1 = 0.0592$ and $wR2 = 0.1336$. The structure is similar to that of $\text{K}_5\text{NaTi}_6\text{Se}_{27}$. It contains infinite one dimensional chains which run parallel to the c -axis and contain Ti^{4+} centers bonded to Se_2^{2-} and Se^{2-} species. The $1/\infty[\text{Ti}_2(\text{Se}_2)_4\text{Se}]^{2-}$ chains pack together in a hexagonal pattern. All of the titanium atoms are surrounded by a pentagonal bipyramid of selenium atoms. The selenium atoms bridge the Ti centers in two different ways. $\text{Na}_2\text{Ti}_2\text{Se}_9$ is a semiconductor with a band gap, E_g , of ~ 1.02 eV. The Raman spectra are also reported.