

NOTIZEN

**Hydroxyamino Sugar Derivatives:
New Glycolipid Analogs**Jean M. J. Tronchet, Daniel Cabrini,
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received May 18, 1992Glycolipid Analogs, Hydroxyamino Sugars,
Non-Ionic Detergents, Reductive Alkylation

Synthetic glycolipid analogs are very useful as nonionic detergents for solubilization of membrane proteins [1]. Although a large number of such detergents were described, very few ones fully satisfy the desirable properties needed for such compounds [2].

Our continuous interest in hydroxyaminosugars [3] lead us to the preparation of some new glycolipid analogs bearing a hydroxylamine moiety [4]. We report hereinafter the preparation of some of these molecules.

Reduction (NaBH_4 , 5 eq.) of the nitron **1** [5] in ethyl alcohol gave the hydroxyamino derivative **2** in acceptable yields (65%). Acetylation of **2** with a large excess of acetic anhydride in pyridine yielded **3** (95%) which gave upon oxidation with *m*-chloroperbenzoic acid (1 eq.) in cold chloroform (4 °C) the epoxyde **4** in 96% yield (Fig. 1). Acidic hydrolysis (1 N HCl, MeOH 1:1) of **2** and **3** failed to give any pure compound. The prepared citronellal derivatives **2**, **3** and **4** are stable thick liquids bearing a pleasant light lemon smell.

Another pathway to glycolipid analogs, involving the hydroxyaminosugar **5** [6] as a starting material was also explored (Fig. 2). Such a cyclopent-

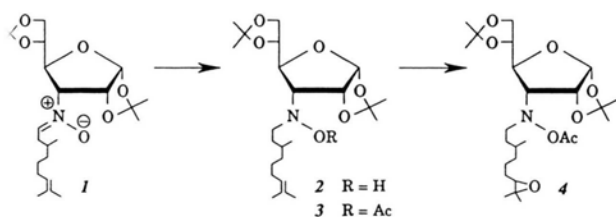


Fig. 1. Synthesis of citronellal derivatives.

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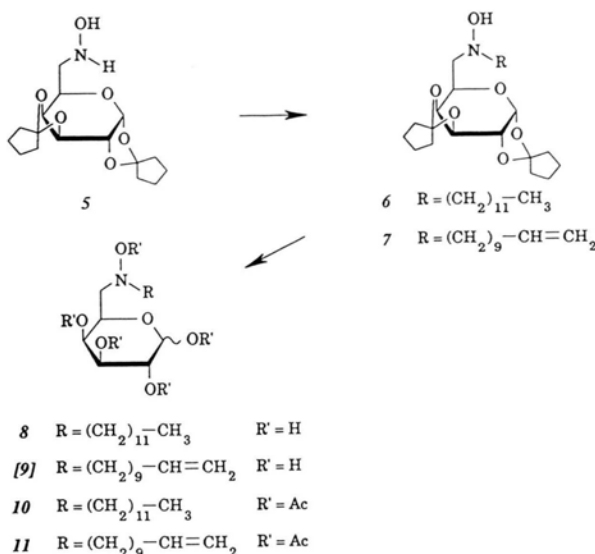
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Fig. 2. Synthesis of dodecanaldehyde and undecenaldehyde derivatives.

Table I. Microanalytical data of newly synthesized compounds.

Nr.	Formula	Mol. weight	Calcd [%]	Found
2	$\text{C}_{22}\text{H}_{39}\text{N}_1\text{O}_6$	413.56	C 63.90	C 64.01
			H 9.51	H 9.72
			N 3.39	N 3.27
3	$\text{C}_{24}\text{H}_{41}\text{N}_1\text{O}_7$	455.60	C 63.27	C 63.16
			H 9.07	H 9.01
			N 3.07	N 3.18
4	$\text{C}_{24}\text{H}_{41}\text{N}_1\text{O}_8$	471.60	C 61.13	C 60.94
			H 8.76	H 8.57
			N 2.97	N 3.00
6	$\text{C}_{28}\text{H}_{49}\text{N}_1\text{O}_6$	495.71	C 67.85	C 67.69
			H 9.96	H 9.93
			N 2.83	N 3.00
7	$\text{C}_{27}\text{H}_{45}\text{N}_1\text{O}_6$	479.66	C 67.61	C 67.58
			H 9.46	H 9.36
			N 2.92	N 3.12
8	$\text{C}_{18}\text{H}_{37}\text{N}_1\text{O}_6$	363.50	C 59.48	C 59.44
			H 10.26	H 10.27
			N 3.85	N 3.89
10	$\text{C}_{28}\text{H}_{47}\text{N}_1\text{O}_{11}$	573.69	C 58.62	C 58.55
			H 8.26	H 8.34
			N 2.44	N 2.55
11	$\text{C}_{27}\text{H}_{42}\text{N}_1\text{O}_{11}$	556.64	C 58.16	C 58.44
			H 7.77	H 7.73
			N 2.51	N 2.71



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tylidene derivative is more sensitive to acidic hydrolysis than its isopropylidene counterpart [7]. Reductive alkylation (NaBH_4 , 5 eq.) in ethyl alcohol of the galactose derivative **5** with dodecanaldehyde (3 eq.) gave the N-dodecyl hydroxyaminosugar **6** in 60% overall yields. **6** was then cleaved (1N HCl, MeOH 1:1) into the unprotected **8** (55%) and acetylated (Ac_2O , pyridine) into the peracetyl glycolipid analog **10** (90%).

The same general procedures, applied to **5** and undecenaldehyde (3 eq.), gave the N-undecenyl

hydroxyaminosugar **7** (55%), then the unprotected **9** which was directly acetylated into the peracetyl glycolipid analog **11** (50% from **7**).

All the newly synthesized compounds **2**, **3**, **4**, **6**, **7**, **8**, **9**, **10** and **11** showed UV, IR, PMR and SM spectra in accordance with their structure [8]. Elementary analysis data [9] are presented in Table I.

Application of the above synthetic route to nucleoside derivatives is presently in progress in our laboratory.

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[8] For the general experimental methods see reference [6]. Details are available from the author (G. Z.-L.).
[9] We are grateful to Dr. H. Eder for performing the microanalyses.