

## The Terbium-Bismuth Alloy System

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The phase diagrams of Tb-Bi has been obtained. A peritectic reaction develops as a result of bismuth addition, the transformation temperature of terbium being lowered by 18°C. Eutec reactions occur at 17 at.%Bi and 1090°C, and at more than 99 at.%Bi and 269°C. There are four compounds in the system: Tb<sub>5</sub>Bi<sub>3</sub>, Tb<sub>4</sub>Bi<sub>3</sub>, TbBi, and TbBi<sub>2</sub>, resulting from peritectic reactions at 1480, 1570, 1770 and 720°C. While TbBi melts congruently at 1920°C. Tb<sub>5</sub>Bi<sub>3</sub> and TbBi were observed to exhibit transformation.

*Key words:* Terbium; Bismuth; Phase Diagram; Alloy.

### 1. Introduction

The purpose of this work was to study phase relationships in the Tb-Bi system and to produce a phase diagram.

Four phases between terbium and bismuth have previously been reported: Tb<sub>5</sub>Bi<sub>3</sub> with the Mn<sub>5</sub>Si<sub>3</sub> structure [1], Tb<sub>5+x</sub>Bi<sub>3</sub> with the Y<sub>5</sub>Bi<sub>3</sub> structure [2], Tb<sub>4</sub>Bi<sub>3</sub> with the anti-Th<sub>3</sub>P<sub>4</sub> structure [3, 4] and TbBi with NaCl structure [5].

### 2. Experimental

#### A) Materials

Terbium distillate TbMD-2 and bismuth of semiconducting purity SCH-0000 were used for the investigation. The major impurities in the terbium are given in Table 1.

Table 1. The impurity concentrations in Terbium Tb MD-2.

Impurity concentration	C	Ca	Dy, Y, Gd	Cu	Fe	Si	H <sub>2</sub>
at %	0.003	0.004	<0.05	<0.01	<0.001	<0.01	0.002

#### B) Preparation of Alloys

25 compositions in the concentration range 0–50 at.%Bi and 15 compositions in the range 50–100 at.%Bi were prepared for the investigation. Each composition was prepared in duplicate or triplicate, and differential thermal analyses (DTA) of the samples were carried out in parallel. The alloys were formed from stoichiometric amounts of the powdered components placed in evacuated quartz tubes which were heated in a resistance furnace. The temperature increase of step-by-step to reach 400°C in 70 h. Homogeneity was obtained by heating the pellets in a crucible with a lid up to a temperature will above the liquidus in an atmosphere of very pure helium. The same technique was used for both the homogeneous annealing and the thermal analysis.

#### C) Examination of Alloys

1. A high temperature differential thermal analyzer was used for the DTA measurements, which were made in very pure helium. The heating and cooling rates were both 30°C/min.

2. X-ray analysis was carried out with Cu K $\alpha$  radiation using a Dron-3 diffractometer.

3. A microscope of “Neofot” 32-type was used for microstructure studies.

### 3. Results and Discussion

An inverse peritectic reaction occurs at the terbium-rich end of the system at 1272°C and at approximately 1 at.%Bi. Evidence for this reaction was obtained from the heating and cooling curves which showed a decrease of approximately 18°C in the  $\alpha$ Tb  $\leftrightarrow$   $\beta$ Tb transformation. This decrease was caused by bismuth additions. A eutectic event occurs on the DTA curve of the sample with a content of 1 at.%Bi. Only two phases were detected by the microscopy and X-ray analyses in the range 0–37.5 at.%Bi, i.e.  $\alpha$ Tb and Tb<sub>5</sub>Bi<sub>3</sub>.

The temperature of the polymorphic transformation in terbium decreased by 18°C on bismuth additions. The eutectic  $e_1$  corresponds to 17 at.%Bi at 1090°C. Tb<sub>5</sub>Bi<sub>3</sub>, being the most terbium-rich phase in the Tb-Bi system, melts incongruently at 1470°C, but is in the DTA curves of the alloys preceded by sharp events at 1400°C. Apparently Tb<sub>5</sub>Bi<sub>3</sub> has a polymorphic transformation. Perhaps, Yoshihara et al. took for Tb<sub>5+x</sub> instead

Table 2. Crystallographic data of the compounds of the alloy system.

Com- pounds	Crystal system	Struc- ture type	Parameters, nm			Ref.
			<i>a</i>	<i>b</i>	<i>c</i>	
$\beta$ -Tb <sub>5</sub> Bi <sub>3</sub> ?	Orthorm	Y <sub>5</sub> Bi <sub>3</sub>	0.8199 0.819	0.9975 0.948	1.1999 1.187	[2] our
$\alpha$ -Tb <sub>5</sub> Bi <sub>3</sub> ?	Hexagon.	Mn <sub>5</sub> Si <sub>3</sub>	0.91006 0.901	– –	0.6365 0.639	[2] our
Tb <sub>4</sub> Bi <sub>3</sub>	Cubic	anti- Th <sub>3</sub> P <sub>4</sub>	0.9328 0.9325 0.930	– – –	– – –	[1] [2] our
TbBi	Cubic	NaCl	0.6277 0.628	– –	– –	[5] our

$\beta$ -Tb<sub>5</sub>Bi<sub>3</sub> [2]. We didn't obtain one-phase samples. X-ray analysis has shown two phases. The lattice parameters of two phases of Tb<sub>5</sub>Bi<sub>3</sub>, have been calculated from several lines and have been coordinated with the data of other authors (Table 2). We have met the problem of polymorphic transformations in Ln<sub>5</sub>Sb<sub>3</sub> in several phase diagrams of heavy REE's.

Tb<sub>4</sub>Bi<sub>3</sub> forms peritectically at 1570°C. Tb-Bi melts congruently at 1920°C and is the biggest melting phase in the system. In DTA curves, Tb-Bi shows a sharp event at 1770°C. DTA and X-ray analyses showed a polymorphic transformation for the first time. However, Hulliger [6] has shown that monoantimonides have a high temperature modification of the CsCl-type.

The DTA curves for alloys containing 52.5–95 at.%Bi exhibit events at 720°C as well as eutectic events. The apex of the Tamman triangle corresponds

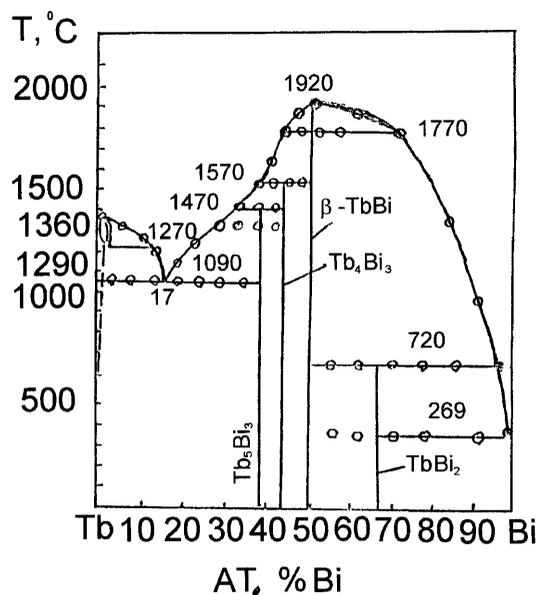


Fig. 1. Phase diagram of Tb-Bi.

to TbBi<sub>2</sub>, which was identified on the basis of DTA and metallographic data without separation into an individual phase. When alloys containing more than 55.5 at.%Bi were heated and cooled, the DTA curves showed peaks near 720°C. At this temperature TbBi<sub>2</sub> is peritectically formed. The structure of TbBi<sub>2</sub> is unknown.

The lattice parameters of the compounds are given in Table 2, and the phase diagram of the Tb-Bi system is shown in Figure 1.

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