

Structure and Phase Transition in $(\text{C}_2\text{H}_5\text{NH}_3)_3\text{Sb}_2\text{Cl}_9 \cdot (\text{C}_2\text{H}_5\text{NH}_3)\text{SbCl}_4$; X-ray, DSC and Dielectric Studies

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The structure of $(\text{C}_2\text{H}_5\text{NH}_3)_3\text{Sb}_2\text{Cl}_9 \cdot (\text{C}_2\text{H}_5\text{NH}_3)\text{SbCl}_4$ at 295 K has been determined. The crystals are orthorhombic, space group $\text{Pna}2_1$ ($a = 16.925(3)$, $b = 24.703(5)$, $c = 7.956(2)$ Å, $V = 3326.4(12)$ Å³, $Z = 4$, $d_c = 2.018$, $d_m = 2.01(1)$ Mg m⁻³). They consist of an anionic sublattice composed of two different polymeric zig-zag chains. One is built of $\text{Sb}_2\text{Cl}_9^{3-}$ units (corner sharing octahedra) and the other one is made of corner sharing SbCl_5^{2-} square pyramids. In the cavities between the polyanionic chains four non-equivalent ethylammonium cations are located. Three of them are disordered. The cations are connected to the anions by weak N-H...Cl hydrogen bonds. A first order phase transition of the order-disorder type was found at 274 K. It was studied by DSC, dielectric and X-ray diffraction methods. The mechanism of the phase transition is attributed to the ordering of at least one of the ethylammonium cations.

Key words: Ethylamine; Chloroantimonate(III); Structure; Phase Transition.