

Studies of Structure and Phase Transition in $[\text{C}(\text{NH}_2)_3]\text{HgBr}_3$ and $[\text{C}(\text{NH}_2)_3]\text{HgI}_3$ by Means of Halogen NQR, ^1H NMR, and Single Crystal X-Ray Diffraction

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The crystal structure of $[\text{C}(\text{NH}_2)_3]\text{HgBr}_3$ was determined at room temperature: monoclinic, space group $\text{C}2/c$, $Z = 4$, $a = 775.0(2)$, $b = 1564.6(2)$, $c = 772.7(2)$ pm, $\beta = 109.12(2)^\circ$. In the crystal, almost planar HgBr_3^- ions are connected via $\text{Hg}\cdots\text{Br}$ bonds, resulting in single chains of trigonal bipyramidal HgBr_5 units which run along the c direction. $[\text{C}(\text{NH}_2)_3]\text{HgI}_3$ was found to be isomorphous with the bromide at room temperature. The temperature dependence of the halogen NQR frequencies ($77 < T/\text{K} < \text{ca. } 380$) and the DTA measurements evidenced no phase transition for the bromide, but a second-order phase transition at (251 ± 1) K (T_{c1}) and a first-order one at (210 ± 1) K (T_{c2}) for the iodide. The transitions at T_{c2} are accompanied with strong supercooling and significant superheating. The room temperature phase (RTP) and the intermediate temperature phase (ITP) of the iodide are characterized by two $^{127}\text{I}_{(m=1/2 \leftrightarrow 3/2)}$ NQR lines which are assigned to the terminal and the bridging I atoms, respectively. There exist three lines in the lowest temperature phase (LTP), indicating that the resonance line of the bridging atom splits into two. The signal intensities of the $^{127}\text{I}_{(m=1/2 \leftrightarrow 3/2)}$ NQR lines in the LTP decrease with decreasing temperature resulting in no detection below ca. 100 K. The $^{127}\text{I}_{(m=1/2 \leftrightarrow 3/2)}$ NQR frequency vs. temperature curves are continuous at T_{c1} , but they are unusual in the LTP. The T_1 vs. T curves of ^1H NMR for the bromide and iodide are explainable by the reorientational motions of the cations about their pseudo three-fold axes. The estimated activation energies of the motions are 35.0 kJ/mol for the bromide, and 24.1, 30.1, and 23.0 kJ/mol for the RTP, ITP, and LTP of the iodide, respectively.

Key words: $[\text{C}(\text{NH}_2)_3]\text{HgX}_3$; Crystal Structure; Phase Transition; NQR; ^1H NMR.