

# Studies of Structure and Phase Transition in $[\text{C}(\text{NH}_2)_3]\text{HgBr}_3$ and $[\text{C}(\text{NH}_2)_3]\text{HgI}_3$ by Means of Halogen NQR, $^1\text{H}$ NMR, and Single Crystal X-Ray Diffraction

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The crystal structure of  $[\text{C}(\text{NH}_2)_3]\text{HgBr}_3$  was determined at room temperature: monoclinic, space group  $\text{C}2/c$ ,  $Z = 4$ ,  $a = 775.0(2)$ ,  $b = 1564.6(2)$ ,  $c = 772.7(2)$  pm,  $\beta = 109.12(2)^\circ$ . In the crystal, almost planar  $\text{HgBr}_3^-$  ions are connected via  $\text{Hg}\cdots\text{Br}$  bonds, resulting in single chains of trigonal bipyramidal  $\text{HgBr}_5$  units which run along the  $c$  direction.  $[\text{C}(\text{NH}_2)_3]\text{HgI}_3$  was found to be isomorphous with the bromide at room temperature. The temperature dependence of the halogen NQR frequencies ( $77 < T/\text{K} < \text{ca. } 380$ ) and the DTA measurements evidenced no phase transition for the bromide, but a second-order phase transition at  $(251 \pm 1)$  K ( $T_{c1}$ ) and a first-order one at  $(210 \pm 1)$  K ( $T_{c2}$ ) for the iodide. The transitions at  $T_{c2}$  are accompanied with strong supercooling and significant superheating. The room temperature phase (RTP) and the intermediate temperature phase (ITP) of the iodide are characterized by two  $^{127}\text{I}_{(m=1/2 \leftrightarrow 3/2)}$  NQR lines which are assigned to the terminal and the bridging I atoms, respectively. There exist three lines in the lowest temperature phase (LTP), indicating that the resonance line of the bridging atom splits into two. The signal intensities of the  $^{127}\text{I}_{(m=1/2 \leftrightarrow 3/2)}$  NQR lines in the LTP decrease with decreasing temperature resulting in no detection below ca. 100 K. The  $^{127}\text{I}_{(m=1/2 \leftrightarrow 3/2)}$  NQR frequency vs. temperature curves are continuous at  $T_{c1}$ , but they are unusual in the LTP. The  $T_1$  vs.  $T$  curves of  $^1\text{H}$  NMR for the bromide and iodide are explainable by the reorientational motions of the cations about their pseudo three-fold axes. The estimated activation energies of the motions are 35.0 kJ/mol for the bromide, and 24.1, 30.1, and 23.0 kJ/mol for the RTP, ITP, and LTP of the iodide, respectively.

*Key words:*  $[\text{C}(\text{NH}_2)_3]\text{HgX}_3$ ; Crystal Structure; Phase Transition; NQR;  $^1\text{H}$  NMR.